

# Benzene, pentafluoroiodo-

<b>Other names:</b>	Iodopentafluorobenzene Pentafluoroiodobenzene Pentafluorophenyl iodide Iodoperfluorobenzene 2,3,4,5,6-Pentafluoroiodobenzene Perfluoroiodobenzene 1,2,3,4,5-Pentafluoro-6-iodo-benzene
<b>Inchi:</b>	InChI=1S/C6F5I/c7-1-2(8)4(10)6(12)5(11)3(1)9
<b>InchiKey:</b>	OPYHNLNYCRZOGY-UHFFFAOYSA-N
<b>Formula:</b>	C6F5I
<b>SMILES:</b>	Fc1c(F)c(F)c(I)c(F)c1F
<b>Mol. weight [g/mol]:</b>	293.96
<b>CAS:</b>	827-15-6

## Physical Properties

Property code	Value	Unit	Source
chl	-2201.00 ± 13.00	kJ/mol	NIST Webbook
ea	1.41 ± 0.11	eV	NIST Webbook
gf	-852.03	kJ/mol	Joback Method
hf	-549.00 ± 13.00	kJ/mol	NIST Webbook
hfus	23.20	kJ/mol	Joback Method
hvap	57.90 ± 0.20	kJ/mol	NIST Webbook
ie	9.54	eV	NIST Webbook
ie	9.54	eV	NIST Webbook
ie	9.50 ± 0.10	eV	NIST Webbook
log10ws	-4.28		Crippen Method
logp	2.987		Crippen Method
mcvol	106.310	ml/mol	McGowan Method
pc	3138.51	kPa	Joback Method
tb	439.00	K	NIST Webbook
tb	434.20	K	NIST Webbook
tc	677.68	K	Joback Method
tf	307.41	K	Joback Method
vc	0.442	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	184.49	J/molxK	477.75	Joback Method
cpg	189.71	J/molxK	511.07	Joback Method
cpg	194.67	J/molxK	544.39	Joback Method
cpg	199.37	J/molxK	577.71	Joback Method
cpg	203.82	J/molxK	611.03	Joback Method
cpg	208.04	J/molxK	644.36	Joback Method
cpg	212.01	J/molxK	677.68	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C827156&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C827156&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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