

Cobalt

Other names:	cobalt element
Inchi:	InChI=1S/Co
InchiKey:	GUTLYIVDDKVIGB-UHFFFAOYSA-N
Formula:	Co
SMILES:	[Co]
Mol. weight [g/mol]:	58.93
CAS:	7440-48-4

Physical Properties

Property code	Value	Unit	Source
affp	742.70	kJ/mol	NIST Webbook
basg	719.80	kJ/mol	NIST Webbook
ea	0.66 ± 0.00	eV	NIST Webbook
ea	0.66 ± 0.00	eV	NIST Webbook
ea	0.66 ± 0.01	eV	NIST Webbook
ie	7.88	eV	NIST Webbook
ie	7.88	eV	NIST Webbook
ie	7.88	eV	NIST Webbook
ie	7.86	eV	NIST Webbook
ie	7.86 ± 0.06	eV	NIST Webbook
ie	7.86	eV	NIST Webbook
ie	7.86	eV	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	ln(Pvp) = A + B/(T + C)
Coeff. A	1.83684e+01
Coeff. B	-4.25454e+04
Coeff. C	-1.06930e+02
Temperature range (K), min.	1790.15
Temperature range (K), max.	3198.15

Sources

Cheméo Relay (1.0):

Thermodynamic excess quantities of ternary Au Co Pd melts by computer-aided analysis of experimental and theoretical study: NIST Webbook

<https://www.doi.org/10.1016/j.tca.2009.05.010>

Thermodynamic studies on LnCoO₃(s) (Ln = Dy, Ho) by solid-state electrochemical cells

<https://www.doi.org/10.1016/j.tca.2015.05.024>

Thermodynamic modeling of the Co Hf system supported by key experiments and first-principles calculations: NIST Webbook

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C7440484&Units=SI>

Low-Temperature Heat Capacities and Standard Molar Enthalpy of Formation of Liquid and Solid Niobium, Zirconium, and Hafnium: NIST Webbook

<https://www.doi.org/10.1021/jc901051z>

Thermodynamic modeling of the Co Hf system supported by key experiments and first-principles calculations: NIST Webbook

<https://www.doi.org/10.1016/j.jct.2014.12.007>

Pressure:

<https://www.doi.org/10.1016/j.jct.2015.02.014>

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Legend

affp:	Proton affinity
basg:	Gas basicity
ea:	Electron affinity
ie:	Ionization energy
vpap:	Vapor pressure

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