

# Furfuryl 2-methyl-2-propenyl sulfide

<b>Inchi:</b>	InChI=1S/C10H14S/c1-9(2)7-11-8-10-5-3-4-6-10/h3-5H,1,6-8H2,2H3
<b>InchiKey:</b>	WPTWMMCUSDCDCL-UHFFFAOYSA-N
<b>Formula:</b>	C10H14S
<b>SMILES:</b>	C=C(C)CSCC1=CC=CC1
<b>Mol. weight [g/mol]:</b>	166.28

## Physical Properties

Property code	Value	Unit	Source
gf	240.28	kJ/mol	Joback Method
hf	92.69	kJ/mol	Joback Method
hfus	18.12	kJ/mol	Joback Method
hvap	45.89	kJ/mol	Joback Method
log10ws	-3.35		Crippen Method
logp	3.182		Crippen Method
mcvol	144.350	ml/mol	McGowan Method
pc	2940.89	kPa	Joback Method
rinpol	1226.00		NIST Webbook
rinpol	1228.00		NIST Webbook
rinpol	1216.00		NIST Webbook
rinpol	1218.00		NIST Webbook
rinpol	1223.00		NIST Webbook
rinpol	1216.00		NIST Webbook
rinpol	1223.00		NIST Webbook
rinpol	1223.00		NIST Webbook
tb	516.79	K	Joback Method
tc	741.94	K	Joback Method
tf	250.32	K	Joback Method
vc	0.545	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	308.66	J/molxK	516.79	Joback Method
cpg	323.91	J/molxK	554.32	Joback Method

cpg	338.20	J/mol×K	591.84	Joback Method
cpg	351.56	J/mol×K	629.37	Joback Method
cpg	364.04	J/mol×K	666.89	Joback Method
cpg	375.70	J/mol×K	704.42	Joback Method
cpg	386.58	J/mol×K	741.94	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R43873&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R43873&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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