

# Diethylmalonic acid, 2,4-dichloro-6-formylphenyl octyl ester

**Inchi:** InChI=1S/C22H30Cl2O5/c1-4-7-8-9-10-11-12-28-20(26)22(5-2,6-3)21(27)29-19-16(15-25)

**InchiKey:** GZSAGXZNRDPRQ-UHFFFAOYSA-N

**Formula:** C22H30Cl2O5

**SMILES:** CCCCCCOC(=O)C(CC)(CC)C(=O)Oc1c(Cl)cc(Cl)cc1C=O

**Mol. weight [g/mol]:** 445.38

## Physical Properties

Property code	Value	Unit	Source
gf	-370.50	kJ/mol	Joback Method
hf	-910.70	kJ/mol	Joback Method
hfus	54.45	kJ/mol	Joback Method
hvap	101.33	kJ/mol	Joback Method
log10ws	-7.46		Crippen Method
logp	6.421		Crippen Method
mcvol	338.010	ml/mol	McGowan Method
pc	1162.45	kPa	Joback Method
rinpol	2779.00		NIST Webbook
rinpol	2779.00		NIST Webbook
tb	1017.25	K	Joback Method
tc	1246.09	K	Joback Method
tf	650.26	K	Joback Method
vc	1.312	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1056.48	J/molxK	1017.25	Joback Method
cpg	1105.47	J/molxK	1207.95	Joback Method
cpg	1098.01	J/molxK	1169.81	Joback Method
cpg	1089.43	J/molxK	1131.67	Joback Method
cpg	1079.68	J/molxK	1093.53	Joback Method
cpg	1068.72	J/molxK	1055.39	Joback Method
cpg	1111.88	J/molxK	1246.09	Joback Method
dvisc	0.0000248	Paxs	1017.25	Joback Method

dvisc	0.0000315	Paxs	956.09	Joback Method
dvisc	0.0000412	Paxs	894.92	Joback Method
dvisc	0.0000562	Paxs	833.75	Joback Method
dvisc	0.0000804	Paxs	772.59	Joback Method
dvisc	0.0001224	Paxs	711.42	Joback Method
dvisc	0.0002016	Paxs	650.26	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370068&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370068&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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