

# 2-(p-Chlorophenyl)-3-methylbutyronitrile

<b>Other names:</b>	Benzeneacetonitrile, 4-chloro-«alpha»-(1-methylethyl)- 2-(4-chlorophenyl)-3-methylbutyronitrile
<b>Inchi:</b>	InChI=1S/C11H12ClN/c1-8(2)11(7-13)9-3-5-10(12)6-4-9/h3-6,8,11H,1-2H3
<b>InchiKey:</b>	RBGSZIRWNWQDOK-UHFFFAOYSA-N
<b>Formula:</b>	C11H12ClN
<b>SMILES:</b>	CC(C)C(C#N)c1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	193.67
<b>CAS:</b>	2012-81-9

## Physical Properties

Property code	Value	Unit	Source
gf	260.89	kJ/mol	Joback Method
hf	93.27	kJ/mol	Joback Method
hfus	16.56	kJ/mol	Joback Method
hvap	57.11	kJ/mol	Joback Method
log10ws	-3.80		Crippen Method
logp	3.603		Crippen Method
mvol	155.710	ml/mol	McGowan Method
pc	2490.03	kPa	Joback Method
tb	621.37	K	Joback Method
tc	854.93	K	Joback Method
tf	317.58	K	Joback Method
vc	0.607	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	359.45	J/mol×K	621.37	Joback Method
cpg	372.26	J/mol×K	660.30	Joback Method
cpg	384.18	J/mol×K	699.22	Joback Method
cpg	395.26	J/mol×K	738.15	Joback Method
cpg	405.53	J/mol×K	777.08	Joback Method
cpg	415.05	J/mol×K	816.01	Joback Method
cpg	423.85	J/mol×K	854.93	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	400.50 ± 1.50	K	0.01	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2012819&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2012819&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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