

# 1,2-Benzenedicarboxaldehyde

<b>Other names:</b>	Phthalic dicarboxaldehyde o-Phthalaldehyde o-Phthalicdicarboxaldehyde Phthaldialdehyde Phthalaldialdehyde o-Phthaldehyde o-Phthaldialdehyde Phthalaldehyde Phthalic aldehyde Phthalic dialdehyde Phthalyldicarboxaldehyde 1,2-Diformylbenzene NSC 13394
<b>Inchi:</b>	InChI=1S/C8H6O2/c9-5-7-3-1-2-4-8(7)6-10/h1-6H
<b>InchiKey:</b>	ZWLUXSQADUDCSB-UHFFFAOYSA-N
<b>Formula:</b>	C8H6O2
<b>SMILES:</b>	O=Cc1ccccc1C=O
<b>Mol. weight [g/mol]:</b>	134.13
<b>CAS:</b>	643-79-8

## Physical Properties

Property code	Value	Unit	Source
gf	-79.78	kJ/mol	Joback Method
hf	-154.55	kJ/mol	Joback Method
hfus	14.71	kJ/mol	Joback Method
hvap	49.78	kJ/mol	Joback Method
log10ws	-1.95		Crippen Method
logp	1.312		Crippen Method
mcvol	102.960	ml/mol	McGowan Method
pc	4316.89	kPa	Joback Method
tb	511.42	K	Joback Method
tc	733.14	K	Joback Method
tf	302.86	K	Joback Method
vc	0.409	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	208.76	J/molxK	511.42	Joback Method
cpg	218.33	J/molxK	548.37	Joback Method
cpg	227.27	J/molxK	585.33	Joback Method
cpg	235.61	J/molxK	622.28	Joback Method
cpg	243.38	J/molxK	659.23	Joback Method
cpg	250.59	J/molxK	696.19	Joback Method
cpg	257.27	J/molxK	733.14	Joback Method
dvisc	0.0025093	Paxs	302.86	Joback Method
dvisc	0.0015427	Paxs	337.62	Joback Method
dvisc	0.0010385	Paxs	372.38	Joback Method
dvisc	0.0007480	Paxs	407.14	Joback Method
dvisc	0.0005673	Paxs	441.90	Joback Method
dvisc	0.0004480	Paxs	476.66	Joback Method
dvisc	0.0003653	Paxs	511.42	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C643798&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C643798&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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