

# Diallyl azelate

<b>Inchi:</b>	InChI=1S/C15H24O4/c1-3-12-18-14(16)10-8-6-5-7-9-11-15(17)19-13-4-2/h3-4H,1-2,5-13
<b>InchiKey:</b>	XZPPRVFUMUEKLN-UHFFFAOYSA-N
<b>Formula:</b>	C15H24O4
<b>SMILES:</b>	C=CCOC(=O)CCCCCCCC(=O)OCC=C
<b>Mol. weight [g/mol]:</b>	268.35

## Physical Properties

Property code	Value	Unit	Source
gf	-216.74	kJ/mol	Joback Method
hf	-591.67	kJ/mol	Joback Method
hfus	37.62	kJ/mol	Joback Method
hvap	65.96	kJ/mol	Joback Method
log10ws	-3.53		Crippen Method
logp	3.175		Crippen Method
mcvol	228.490	ml/mol	McGowan Method
pc	1623.29	kPa	Joback Method
rinsol	1808.00		NIST Webbook
tb	688.54	K	Joback Method
tc	868.31	K	Joback Method
tf	399.61	K	Joback Method
vc	0.885	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	629.63	J/molxK	688.54	Joback Method
cpg	697.57	J/molxK	838.35	Joback Method
cpg	685.43	J/molxK	808.39	Joback Method
cpg	672.58	J/molxK	778.42	Joback Method
cpg	659.01	J/molxK	748.46	Joback Method
cpg	644.69	J/molxK	718.50	Joback Method
cpg	709.00	J/molxK	868.31	Joback Method
dvisc	0.0001171	Paxs	688.54	Joback Method
dvisc	0.0001511	Paxs	640.38	Joback Method

dvisc	0.0002031	Paxs	592.23	Joback Method
dvisc	0.0002876	Paxs	544.07	Joback Method
dvisc	0.0004358	Paxs	495.92	Joback Method
dvisc	0.0007223	Paxs	447.76	Joback Method
dvisc	0.0013518	Paxs	399.61	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R542293&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R542293&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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