

# 2-(4-Methyl-pentyl)-3-decyl-oxirane

<b>Other names:</b>	2-(5-Methyl-pentyl)-3-decyl-oxirane
<b>Inchi:</b>	InChI=1S/C18H36O/c1-4-5-6-7-8-9-10-11-14-17-18(19-17)15-12-13-16(2)3/h16-18H,4-1
<b>InchiKey:</b>	MYAXSLNSJQGASK-UHFFFAOYSA-N
<b>Formula:</b>	C18H36O
<b>SMILES:</b>	CCCCCCCCCCC1OC1CCCC(C)C
<b>Mol. weight [g/mol]:</b>	268.48

## Physical Properties

Property code	Value	Unit	Source
gf	65.16	kJ/mol	Joback Method
hf	-499.67	kJ/mol	Joback Method
hfus	46.04	kJ/mol	Joback Method
hvap	59.39	kJ/mol	Joback Method
log10ws	-6.32		Crippen Method
logp	6.111		Crippen Method
mcvol	259.490	ml/mol	McGowan Method
pc	1231.15	kPa	Joback Method
rinpol	1896.00		NIST Webbook
tb	639.82	K	Joback Method
tc	809.59	K	Joback Method
tf	317.89	K	Joback Method
vc	1.014	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	751.06	J/molxK	639.82	Joback Method
cpg	845.42	J/molxK	781.29	Joback Method
cpg	828.26	J/molxK	753.00	Joback Method
cpg	810.28	J/molxK	724.70	Joback Method
cpg	791.44	J/molxK	696.41	Joback Method
cpg	771.71	J/molxK	668.11	Joback Method
cpg	861.80	J/molxK	809.59	Joback Method
dvisc	0.0003148	Paxs	639.82	Joback Method

dvisc	0.0003905	Paxs	586.16	Joback Method
dvisc	0.0005059	Paxs	532.51	Joback Method
dvisc	0.0006946	Paxs	478.85	Joback Method
dvisc	0.0010331	Paxs	425.20	Joback Method
dvisc	0.0017232	Paxs	371.54	Joback Method
dvisc	0.0034163	Paxs	317.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R489464&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R489464&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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