

1-Pentene, 2,3,3-trimethyl-

Other names:	2,3,3-Trimethyl-1-pentene 2,3,3-Trimethylpent-1-ene
Inchi:	InChI=1S/C8H16/c1-6-8(4,5)7(2)3/h2,6H2,1,3-5H3
InchiKey:	TUSBCMPNIOJUBX-UHFFFAOYSA-N
Formula:	C8H16
SMILES:	C=C(C)C(C)(C)CC
Mol. weight [g/mol]:	112.21
CAS:	560-23-6

Physical Properties

Property code	Value	Unit	Source
gf	98.61	kJ/mol	Joback Method
hf	-101.56	kJ/mol	Joback Method
hfus	6.47	kJ/mol	Joback Method
hvap	38.50	kJ/mol	NIST Webbook
log10ws	-2.78		Crippen Method
logp	2.999		Crippen Method
mcvol	119.280	ml/mol	McGowan Method
pc	2712.67	kPa	Joback Method
rinpol	738.00		NIST Webbook
rinpol	743.00		NIST Webbook
rinpol	743.00		NIST Webbook
rinpol	734.00		NIST Webbook
rinpol	730.50		NIST Webbook
rinpol	734.00		NIST Webbook
rinpol	743.30		NIST Webbook
rinpol	743.30		NIST Webbook
rinpol	734.00		NIST Webbook
rinpol	735.00		NIST Webbook
rinpol	735.00		NIST Webbook
rinpol	730.30		NIST Webbook
rinpol	743.00		NIST Webbook
tb	381.46 ± 0.25	K	NIST Webbook
tb	381.40 ± 1.00	K	NIST Webbook
tb	381.55 ± 0.30	K	NIST Webbook
tb	391.65 ± 0.30	K	NIST Webbook
tc	557.27	K	Joback Method

tf	204.00 ± 3.00	K	NIST Webbook
vc	0.455	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	218.61	J/mol×K	375.77	Joback Method
cpg	232.95	J/mol×K	406.02	Joback Method
cpg	246.58	J/mol×K	436.27	Joback Method
cpg	259.50	J/mol×K	466.52	Joback Method
cpg	271.77	J/mol×K	496.77	Joback Method
cpg	283.40	J/mol×K	527.02	Joback Method
cpg	294.42	J/mol×K	557.27	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.36323e+01
Coeff. B	-2.87503e+03
Coeff. C	-6.25060e+01
Temperature range (K), min.	277.95
Temperature range (K), max.	408.03

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C560236&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
KDB:	https://www.thermo.com/files/research/kdb/mol/mol327.mol

Legend

cp_g:	Ideal gas heat capacity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
ri_{npol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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