

# Glutaric acid, 2,6-dimethoxyphenyl propyl ester

Inchi:	InChI=1S/C16H22O6/c1-4-11-21-14(17)9-6-10-15(18)22-16-12(19-2)7-5-8-13(16)20-3/h5
InchiKey:	SHMOVALCIDUASW-UHFFFAOYSA-N
Formula:	C16H22O6
SMILES:	CCCOC(=O)CCCC(=O)Oc1c(OC)cccc1OC
Mol. weight [g/mol]:	310.34

## Physical Properties

Property code	Value	Unit	Source
gf	-500.85	kJ/mol	Joback Method
hf	-914.02	kJ/mol	Joback Method
hfus	38.41	kJ/mol	Joback Method
hvap	77.94	kJ/mol	Joback Method
log10ws	-3.40		Crippen Method
logp	2.733		Crippen Method
mcvol	239.160	ml/mol	McGowan Method
pc	1743.37	kPa	Joback Method
rinpol	2299.00		NIST Webbook
rinpol	2299.00		NIST Webbook
tb	799.54	K	Joback Method
tc	1001.20	K	Joback Method
tf	510.32	K	Joback Method
vc	0.907	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	703.22	J/molxK	799.54	Joback Method
cpg	763.73	J/molxK	967.59	Joback Method
cpg	753.82	J/molxK	933.98	Joback Method
cpg	742.79	J/molxK	900.37	Joback Method
cpg	730.67	J/molxK	866.76	Joback Method
cpg	717.47	J/molxK	833.15	Joback Method
cpg	772.51	J/molxK	1001.20	Joback Method
dvisc	0.0000545	Paxs	799.54	Joback Method

dvisc	0.0000678	Paxs	751.34	Joback Method
dvisc	0.0000870	Paxs	703.13	Joback Method
dvisc	0.0001158	Paxs	654.93	Joback Method
dvisc	0.0001612	Paxs	606.73	Joback Method
dvisc	0.0002376	Paxs	558.52	Joback Method
dvisc	0.0003770	Paxs	510.32	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358704&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358704&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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