

# Naphthalene, 2-nitro-

<b>Other names:</b>	2-Nitronaphthalene «beta»-Nitronaphthalene Â«betaÂ»-Nitronaphthalene
<b>Inchi:</b>	InChI=1S/C10H7NO2/c12-11(13)10-6-5-8-3-1-2-4-9(8)7-10/h1-7H
<b>InchiKey:</b>	ZJYJZEAJZXVAMF-UHFFFAOYSA-N
<b>Formula:</b>	C10H7NO2
<b>SMILES:</b>	O=[N+](O-)c1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	173.17
<b>CAS:</b>	581-89-5

## Physical Properties

Property code	Value	Unit	Source
ea	1.24 ± 0.05	eV	NIST Webbook
ea	1.18 ± 0.10	eV	NIST Webbook
gf	278.30	kJ/mol	Joback Method
hf	155.64	kJ/mol	Joback Method
hfus	23.69	kJ/mol	Joback Method
hvap	59.02	kJ/mol	Joback Method
ie	8.67 ± 0.01	eV	NIST Webbook
ie	8.63	eV	NIST Webbook
log10ws	-3.82		Aqueous Solubility Prediction Method
logp	2.748		Crippen Method
mvol	125.960	ml/mol	McGowan Method
pc	3881.95	kPa	Joback Method
rinpol	280.06		NIST Webbook
rinpol	1635.00		NIST Webbook
rinpol	1672.00		NIST Webbook
rinpol	1631.00		NIST Webbook
rinpol	1631.00		NIST Webbook
rinpol	279.20		NIST Webbook
rinpol	1639.00		NIST Webbook
rinpol	1631.00		NIST Webbook
rinpol	280.32		NIST Webbook
rinpol	275.21		NIST Webbook
rinpol	1639.00		NIST Webbook
rinpol	278.91		NIST Webbook

rmpol	280.63		NIST Webbook
rmpol	279.20		NIST Webbook
rmpol	1631.00		NIST Webbook
rmpol	280.32		NIST Webbook
rmpol	1635.00		NIST Webbook
rmpol	1656.00		NIST Webbook
rmpol	1656.00		NIST Webbook
tb	630.68	K	Joback Method
tc	898.12	K	Joback Method
tf	417.71	K	Joback Method
vc	0.491	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	297.11	J/mol×K	630.68	Joback Method
cpg	308.94	J/mol×K	675.25	Joback Method
cpg	319.67	J/mol×K	719.83	Joback Method
cpg	329.43	J/mol×K	764.40	Joback Method
cpg	338.30	J/mol×K	808.97	Joback Method
cpg	346.42	J/mol×K	853.54	Joback Method
cpg	353.89	J/mol×K	898.12	Joback Method
hfust	14.50	kJ/mol	348.20	NIST Webbook

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C581895&Units=SI>

## Legend

**cpg:** Ideal gas heat capacity

**ea:** Electron affinity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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