

# Benzoyl chloride, 4-chloro-

|                             |   |
|-----------------------------|---|
| <b>Other names:</b>         | 4-Chlorobenzoic acid chloride<br>4-Chlorobenzoyl chloride<br>Benzoyl chloride, p-chloro-<br>p-Chlorobenzoyl chloride<br>para-Chlorobenzoyl chloride |
| <b>Inchi:</b>               | InChI=1S/C7H4Cl2O/c8-6-3-1-5(2-4-6)7(9)10/h1-4H   |
| <b>InchiKey:</b>            | RKIDDEGICSMIJA-UHFFFAOYSA-N   |
| <b>Formula:</b>             | C7H4Cl2O  |
| <b>SMILES:</b>              | O=C(Cl)c1ccc(Cl)cc1   |
| <b>Mol. weight [g/mol]:</b> | 175.01  |
| <b>CAS:</b>                 | 122-01-0  |

## Physical Properties

| Property code | Value          | Unit                 | Source         |
|---------------|----------------|----------------------|----------------|
| gf            | -41.94         | kJ/mol               | Joback Method  |
| hf            | -128.90 ± 4.30 | kJ/mol               | NIST Webbook   |
| hfl           | -190.90 ± 0.88 | kJ/mol               | NIST Webbook   |
| hfl           | -191.70 ± 1.10 | kJ/mol               | NIST Webbook   |
| hfus          | 17.53          | kJ/mol               | Joback Method  |
| hvap          | 49.63          | kJ/mol               | Joback Method  |
| ie            | 9.58 ± 0.03    | eV                   | NIST Webbook   |
| log10ws       | -3.05          |                      | Crippen Method |
| logp          | 2.719          |                      | Crippen Method |
| mcvol         | 111.780        | ml/mol               | McGowan Method |
| pc            | 3980.54        | kPa                  | Joback Method  |
| tb            | 495.20         | K                    | NIST Webbook   |
| tc            | 758.29         | K                    | Joback Method  |
| tf            | 317.36         | K                    | Joback Method  |
| vc            | 0.423          | m <sup>3</sup> /kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 239.57 | J/mol×K | 758.29          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 233.90    | J/molxK | 718.56 | Joback Method |
| cpg   | 227.70    | J/molxK | 678.84 | Joback Method |
| cpg   | 220.95    | J/molxK | 639.12 | Joback Method |
| cpg   | 213.61    | J/molxK | 599.40 | Joback Method |
| cpg   | 205.65    | J/molxK | 559.67 | Joback Method |
| cpg   | 197.03    | J/molxK | 519.95 | Joback Method |
| dvisc | 0.0021445 | Paxs    | 317.36 | Joback Method |
| dvisc | 0.0003235 | Paxs    | 519.95 | Joback Method |
| dvisc | 0.0003974 | Paxs    | 486.19 | Joback Method |
| dvisc | 0.0005035 | Paxs    | 452.42 | Joback Method |
| dvisc | 0.0006626 | Paxs    | 418.66 | Joback Method |
| dvisc | 0.0009150 | Paxs    | 384.89 | Joback Method |
| dvisc | 0.0013446 | Paxs    | 351.12 | Joback Method |
| hvapt | 55.70     | kJ/mol  | 381.00 | NIST Webbook  |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.56902e+01                   |
| Coeff. B                    | -5.77711e+03                  |
| Temperature range (K), min. | 370.00                        |
| Temperature range (K), max. | 556.63                        |

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C122010&Units=SI>

**The Yaws Handbook of Vapor Pressure:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                                   |
| <b>dvisc:</b>   | Dynamic viscosity   |
| <b>gf:</b>      | Standard Gibbs free energy of formation                   |
| <b>hf:</b>      | Enthalpy of formation at standard conditions              |
| <b>hfl:</b>     | Liquid phase enthalpy of formation at standard conditions |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions                 |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions           |
| <b>hvapt:</b>   | Enthalpy of vaporization at a given temperature           |
| <b>ie:</b>      | Ionization energy   |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l                        |
| <b>logp:</b>    | Octanol/Water partition coefficient                       |
| <b>mcvol:</b>   | McGowan's characteristic volume                           |
| <b>pc:</b>      | Critical Pressure   |
| <b>pvap:</b>    | Vapor pressure  |
| <b>tb:</b>      | Normal Boiling Point Temperature                          |
| <b>tc:</b>      | Critical Temperature                                      |
| <b>tf:</b>      | Normal melting (fusion) point                             |
| <b>vc:</b>      | Critical Volume   |

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