

Sebacic acid, decyl pentafluorobenzyl ester

Inchi:	InChI=1S/C27H39F5O4/c1-2-3-4-5-6-9-12-15-18-35-21(33)16-13-10-7-8-11-14-17-22(34)
InchiKey:	GHDGOTYHUHBNED-UHFFFAOYSA-N
Formula:	C27H39F5O4
SMILES:	CCCCCCCCCOC(=O)CCCCCCCC(=O)OCc1c(F)c(F)c(F)c(F)c1F
Mol. weight [g/mol]:	522.59

Physical Properties

Property code	Value	Unit	Source
gf	-1201.17	kJ/mol	Joback Method
hf	-1891.58	kJ/mol	Joback Method
hfus	78.76	kJ/mol	Joback Method
hvap	95.51	kJ/mol	Joback Method
log10ws	-10.11		Crippen Method
logp	8.230		Crippen Method
mvol	391.260	ml/mol	McGowan Method
pc	735.62	kPa	Joback Method
rinpol	3073.00		NIST Webbook
tb	1017.67	K	Joback Method
tc	1269.31	K	Joback Method
tf	630.34	K	Joback Method
vc	1.577	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1337.16	J/mol×K	1017.67	Joback Method
cpg	1355.11	J/mol×K	1059.61	Joback Method
cpg	1370.92	J/mol×K	1101.55	Joback Method
cpg	1384.63	J/mol×K	1143.49	Joback Method
cpg	1396.30	J/mol×K	1185.43	Joback Method
cpg	1405.97	J/mol×K	1227.37	Joback Method
cpg	1413.69	J/mol×K	1269.31	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U354905&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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