

# Methyl palustrate

**Other names:**

1-Phenanthrenecarboxylic acid,  
1,2,3,4,4a,5,6,9,10,10a-decahydro-1,4a-dimethyl-7-(1-methylethyl)-, methyl ester,  
Pectocarpa-8,13-dien-15-oic acid, 1,3-isopropyl-, methyl ester  
[1R-(1«alpha»,4a«beta»,10a«alpha»)]-  
Palastric acid, methyl ester

**Inchi:** InChI=1S/C21H32O2/c1-14(2)15-7-9-17-16(13-15)8-10-18-20(17,3)11-6-12-21(18,4)19(2)**InchiKey:** MEQSJWRGVQAVJY-MNLRITNHSA-N**Formula:** C21H32O2**SMILES:** COC(=O)C1(C)CCCC2(C)C3=C(C=C(C(C)C)CC3)CCC12**Mol. weight [g/mol]:** 316.48**CAS:** 3310-94-9

## Physical Properties

Property code	Value	Unit	Source
gf	31.38	kJ/mol	Joback Method
hf	-427.62	kJ/mol	Joback Method
hfus	22.00	kJ/mol	Joback Method
hvap	71.97	kJ/mol	Joback Method
log10ws	-5.90		Crippen Method
logp	5.439		Crippen Method
mcvol	273.010	ml/mol	McGowan Method
pc	1540.29	kPa	Joback Method
rinpol	2271.00		NIST Webbook
rinpol	2284.00		NIST Webbook
rinpol	2271.00		NIST Webbook
ripol	2876.00		NIST Webbook
ripol	2875.00		NIST Webbook
ripol	2875.00		NIST Webbook
tb	811.04	K	Joback Method
tc	1043.59	K	Joback Method
tf	506.69	K	Joback Method
vc	1.028	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	883.85	J/mol×K	811.04	Joback Method
cpg	908.00	J/mol×K	849.80	Joback Method
cpg	931.89	J/mol×K	888.56	Joback Method
cpg	955.82	J/mol×K	927.31	Joback Method
cpg	980.09	J/mol×K	966.07	Joback Method
cpg	1004.99	J/mol×K	1004.83	Joback Method
cpg	1030.82	J/mol×K	1043.59	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3310949&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3310949&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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