

# Diethylmalonic acid, hexyl 2,4,4-trimethylpentyl ester

Inchi:	InChI=1S/C21H40O4/c1-8-11-12-13-14-24-18(22)21(9-2,10-3)19(23)25-16-17(4)15-20(5)
InchiKey:	HERKBCPQYFCDHL-UHFFFAOYSA-N
Formula:	C21H40O4
SMILES:	CCCCCOC(=O)C(CC)(CC)C(=O)OCC(C)CC(C)(C)C
Mol. weight [g/mol]:	356.54

## Physical Properties

Property code	Value	Unit	Source
gf	-338.66	kJ/mol	Joback Method
hf	-989.15	kJ/mol	Joback Method
hfus	37.37	kJ/mol	Joback Method
hvap	77.67	kJ/mol	Joback Method
log10ws	-5.61		Crippen Method
logp	5.532		Crippen Method
mcvol	321.630	ml/mol	McGowan Method
pc	1043.27	kPa	Joback Method
rinsol	2012.00		NIST Webbook
tb	825.56	K	Joback Method
tc	1017.59	K	Joback Method
tf	460.59	K	Joback Method
vc	1.232	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1032.28	J/molxK	825.56	Joback Method
cpg	1115.03	J/molxK	985.59	Joback Method
cpg	1100.54	J/molxK	953.58	Joback Method
cpg	1085.07	J/molxK	921.58	Joback Method
cpg	1068.57	J/molxK	889.57	Joback Method
cpg	1050.99	J/molxK	857.57	Joback Method
cpg	1128.59	J/molxK	1017.59	Joback Method
dvisc	0.0000268	Paxs	825.56	Joback Method
dvisc	0.0000376	Paxs	764.73	Joback Method

dvisc	0.0000562	Paxs	703.90	Joback Method
dvisc	0.0000905	Paxs	643.08	Joback Method
dvisc	0.0001609	Paxs	582.25	Joback Method
dvisc	0.0003271	Paxs	521.42	Joback Method
dvisc	0.0008026	Paxs	460.59	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369478&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369478&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/19-546-5/Diethylmalonic-acid-hexyl-2-4-4-trimethylpentyl-ester.pdf>

Generated by Cheméo on 2025-12-05 15:14:17.707671378 +0000 UTC m=+4695855.237712037.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.