

# Butanoic acid, 3-chloro, 1-methylpropyl ester

Inchi:	InChI=1S/C8H15ClO2/c1-4-7(3)11-8(10)5-6(2)9/h6-7H,4-5H2,1-3H3
InchiKey:	CDLRHJQHVUEVET-UHFFFAOYSA-N
Formula:	C8H15ClO2
SMILES:	CCC(C)OC(=O)CC(C)Cl
Mol. weight [g/mol]:	178.66

## Physical Properties

Property code	Value	Unit	Source
gf	-234.25	kJ/mol	Joback Method
hf	-479.55	kJ/mol	Joback Method
hfus	16.41	kJ/mol	Joback Method
hvap	46.17	kJ/mol	Joback Method
log10ws	-2.41		Crippen Method
logp	2.345		Crippen Method
mcvol	143.260	ml/mol	McGowan Method
pc	2600.43	kPa	Joback Method
ripol	1082.00		NIST Webbook
ripol	1077.00		NIST Webbook
ripol	1089.00		NIST Webbook
ripol	1065.00		NIST Webbook
ripol	1070.00		NIST Webbook
ripol	1460.00		NIST Webbook
ripol	1442.00		NIST Webbook
ripol	1439.00		NIST Webbook
ripol	1434.00		NIST Webbook
tb	495.28	K	Joback Method
tc	683.89	K	Joback Method
tf	252.00	K	Joback Method
vc	0.544	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	308.11	J/mol×K	495.28	Joback Method

cpg	320.63	J/mol×K	526.72	Joback Method
cpg	332.62	J/mol×K	558.15	Joback Method
cpg	344.09	J/mol×K	589.59	Joback Method
cpg	355.06	J/mol×K	621.02	Joback Method
cpg	365.51	J/mol×K	652.46	Joback Method
cpg	375.47	J/mol×K	683.89	Joback Method
dvisc	0.0061890	Paxs	252.00	Joback Method
dvisc	0.0024580	Paxs	292.55	Joback Method
dvisc	0.0012223	Paxs	333.09	Joback Method
dvisc	0.0007073	Paxs	373.64	Joback Method
dvisc	0.0004556	Paxs	414.19	Joback Method
dvisc	0.0003174	Paxs	454.73	Joback Method
dvisc	0.0002346	Paxs	495.28	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R28764&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R28764&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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