

# p-mentha-1,3-dien-8-ol

<b>Inchi:</b>	InChI=1S/C10H16O/c1-8-4-6-9(7-5-8)10(2,3)11/h4,6,11H,5,7H2,1-3H3
<b>InchiKey:</b>	KZBBYHLCBRURPA-UHFFFAOYSA-N
<b>Formula:</b>	C10H16O
<b>SMILES:</b>	CC1=CC=C(C(C)(C)O)CC1
<b>Mol. weight [g/mol]:</b>	152.23

## Physical Properties

Property code	Value	Unit	Source
gf	-27.84	kJ/mol	Joback Method
hf	-243.43	kJ/mol	Joback Method
hfus	10.76	kJ/mol	Joback Method
hvap	55.88	kJ/mol	Joback Method
log10ws	-2.99		Crippen Method
logp	2.424		Crippen Method
mcvol	138.170	ml/mol	McGowan Method
pc	3131.49	kPa	Joback Method
rinsol	1172.00		NIST Webbook
tb	549.65	K	Joback Method
tc	752.81	K	Joback Method
tf	303.88	K	Joback Method
vc	0.509	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	333.67	J/molxK	549.65	Joback Method
cpg	347.82	J/molxK	583.51	Joback Method
cpg	361.11	J/molxK	617.37	Joback Method
cpg	373.59	J/molxK	651.23	Joback Method
cpg	385.30	J/molxK	685.09	Joback Method
cpg	396.29	J/molxK	718.95	Joback Method
cpg	406.59	J/molxK	752.81	Joback Method
dvisc	0.0118833	Paxs	303.88	Joback Method
dvisc	0.0032840	Paxs	344.84	Joback Method

dvisc	0.0011925	Paxs	385.80	Joback Method
dvisc	0.0005260	Paxs	426.76	Joback Method
dvisc	0.0002678	Paxs	467.73	Joback Method
dvisc	0.0001520	Paxs	508.69	Joback Method
dvisc	0.0000939	Paxs	549.65	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R219240&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R219240&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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