

# Perfluoro-pentan ethioic acid S-ethyl ester

**Inchi:** InChI=1S/C7H5F9OS/c1-2-18-3(17)4(8,9)5(10,11)6(12,13)7(14,15)16/h2H2,1H3  
**InchiKey:** VIZPJQZIXDOYQY-UHFFFAOYSA-N  
**Formula:** C7H5F9OS  
**SMILES:** CCSC(=O)C(F)(F)C(F)(F)C(F)(F)C(F)(F)F  
**Mol. weight [g/mol]:** 308.17

## Physical Properties

Property code	Value	Unit	Source
gf	-1829.67	kJ/mol	Joback Method
hf	-2058.51	kJ/mol	Joback Method
hfus	17.68	kJ/mol	Joback Method
hvap	32.20	kJ/mol	Joback Method
log10ws	-4.02		Crippen Method
logp	3.734		Crippen Method
mcvol	143.340	ml/mol	McGowan Method
pc	2185.64	kPa	Joback Method
ripol	773.00		NIST Webbook
ripol	766.00		NIST Webbook
ripol	752.00		NIST Webbook
ripol	757.00		NIST Webbook
ripol	880.00		NIST Webbook
ripol	821.00		NIST Webbook
ripol	855.00		NIST Webbook
ripol	797.00		NIST Webbook
tb	462.72	K	Joback Method
tc	623.33	K	Joback Method
tf	267.97	K	Joback Method
vc	0.606	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.75	J/molxK	462.72	Joback Method
cpg	361.99	J/molxK	489.49	Joback Method

cpg	372.39	J/mol×K	516.26	Joback Method
cpg	382.00	J/mol×K	543.03	Joback Method
cpg	390.86	J/mol×K	569.80	Joback Method
cpg	399.00	J/mol×K	596.56	Joback Method
cpg	406.49	J/mol×K	623.33	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R183750&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R183750&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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