

# 2-Nitro-«alpha»,«alpha»,«alpha»-trifluorotoluene

<b>Other names:</b>	o-Nitrobenzotrifluoride 2-Nitrobenzotrifluoride Benzene, 1-nitro-2-(trifluoromethyl)- Toluene, 2-nitro-«alpha»,«alpha»,«alpha»-trifluoro- o-(Trifluoromethyl)nitrobenzene Toluene, «alpha»,«alpha»,«alpha»-trifluoro-o-nitro- 1-Nitro-2-(trifluoromethyl)benzene «alpha»,«alpha»,«alpha»-Trifluoro-o-nitrotoluene 2-(Trifluoromethyl)nitrobenzene o-Nitro-«alpha»,«alpha»,«alpha»-trifluorotoluene O-nitro-alpha,alpha,alpha-trifluorotoluene
<b>Inchi:</b>	InChI=1S/C7H4F3NO2/c8-7(9,10)5-3-1-2-4-6(5)11(12)13/h1-4H
<b>InchiKey:</b>	NDZJSUCUYPZXPR-UHFFFAOYSA-N
<b>Formula:</b>	C7H4F3NO2
<b>SMILES:</b>	O=[N+](=O)[O-]c1ccccc1C(F)(F)F
<b>Mol. weight [g/mol]:</b>	191.11
<b>CAS:</b>	384-22-5

## Physical Properties

Property code	Value	Unit	Source
ea	1.33 ± 0.10	eV	NIST Webbook
gf	-435.20	kJ/mol	Joback Method
hf	-570.59	kJ/mol	Joback Method
hfus	20.73	kJ/mol	Joback Method
hvap	46.96	kJ/mol	Joback Method
log10ws	-3.23		Crippen Method
logp	2.614		Crippen Method
mcvol	108.460	ml/mol	McGowan Method
pc	3551.53	kPa	Joback Method
tb	537.64	K	Joback Method
tc	764.81	K	Joback Method
tf	355.39	K	Joback Method
vc	0.445	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	250.24	J/mol×K	537.64	Joback Method
cpg	260.42	J/mol×K	575.50	Joback Method
cpg	269.71	J/mol×K	613.36	Joback Method
cpg	278.17	J/mol×K	651.23	Joback Method
cpg	285.87	J/mol×K	689.09	Joback Method
cpg	292.87	J/mol×K	726.95	Joback Method
cpg	299.22	J/mol×K	764.81	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	377.70	K	2.70	NIST Webbook

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C384225&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C384225&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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