

# Benzene, 1-butyl-4-ethyl

<b>Other names:</b>	1-Ethyl-4-n-Butylbenzene
<b>Inchi:</b>	InChI=1S/C12H18/c1-3-5-6-12-9-7-11(4-2)8-10-12/h7-10H,3-6H2,1-2H3
<b>InchiKey:</b>	UPYBAFFRUZTWFA-UHFFFAOYSA-N
<b>Formula:</b>	C12H18
<b>SMILES:</b>	CCCCc1ccc(CC)cc1
<b>Mol. weight [g/mol]:</b>	162.27

## Physical Properties

Property code	Value	Unit	Source
gf	152.94	kJ/mol	Joback Method
hf	-65.95	kJ/mol	Joback Method
hfus	20.49	kJ/mol	Joback Method
hvap	45.24	kJ/mol	Joback Method
log10ws	-3.91		Crippen Method
logp	3.592		Crippen Method
mcvol	156.180	ml/mol	McGowan Method
pc	2370.28	kPa	Joback Method
rinpol	1231.00		NIST Webbook
rinpol	1232.00		NIST Webbook
rinpol	1232.00		NIST Webbook
rinpol	1227.00		NIST Webbook
rinpol	1231.00		NIST Webbook
rinpol	1236.00		NIST Webbook
rinpol	1227.00		NIST Webbook
ripol	1440.00		NIST Webbook
ripol	1470.30		NIST Webbook
ripol	1439.50		NIST Webbook
ripol	1440.00		NIST Webbook
ripol	1440.00		NIST Webbook
ripol	1470.30		NIST Webbook
tb	505.62	K	Joback Method
tc	706.05	K	Joback Method
tf	263.94	K	Joback Method
vc	0.600	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	346.19	J/molxK	505.62	Joback Method
cpg	420.54	J/molxK	672.64	Joback Method
cpg	407.24	J/molxK	639.24	Joback Method
cpg	393.19	J/molxK	605.83	Joback Method
cpg	378.35	J/molxK	572.43	Joback Method
cpg	362.69	J/molxK	539.02	Joback Method
cpg	433.10	J/molxK	706.05	Joback Method
dvisc	0.0002000	Paxs	505.62	Joback Method
dvisc	0.0002551	Paxs	465.34	Joback Method
dvisc	0.0003409	Paxs	425.06	Joback Method
dvisc	0.0004839	Paxs	384.78	Joback Method
dvisc	0.0007455	Paxs	344.50	Joback Method
dvisc	0.0012879	Paxs	304.22	Joback Method
dvisc	0.0026290	Paxs	263.94	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R12462&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R12462&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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