

# 2-Cyclohexen-1-ol

<b>Other names:</b>	Cyclohexen-3-ol 1-Cyclohexen-3-ol 2-Cyclohexenol 3-Hydroxycyclohexene 2-Cyclohexene-1-ol cyclohex-2-ene-1-ol
<b>Inchi:</b>	InChI=1S/C6H10O/c7-6-4-2-1-3-5-6/h2,4,6-7H,1,3,5H2
<b>InchiKey:</b>	PQANGXXSEABURG-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O
<b>SMILES:</b>	OC1C=CCCC1
<b>Mol. weight [g/mol]:</b>	98.14
<b>CAS:</b>	822-67-3

## Physical Properties

Property code	Value	Unit	Source
gf	-82.77	kJ/mol	Joback Method
hf	-207.30	kJ/mol	Joback Method
hfus	8.44	kJ/mol	Joback Method
hvap	46.35	kJ/mol	Joback Method
log10ws	-1.46		Crippen Method
logp	1.087		Crippen Method
mcvol	86.110	ml/mol	McGowan Method
pc	4684.89	kPa	Joback Method
rinpol	887.00		NIST Webbook
rinpol	883.00		NIST Webbook
rinpol	890.00		NIST Webbook
rinpol	890.00		NIST Webbook
rinpol	887.00		NIST Webbook
rinpol	887.00		NIST Webbook
rinpol	887.00		NIST Webbook
rinpol	880.00		NIST Webbook
ripol	1479.00		NIST Webbook
ripol	1481.00		NIST Webbook
ripol	1427.00		NIST Webbook
ripol	1471.00		NIST Webbook
ripol	1471.00		NIST Webbook
ripol	1473.00		NIST Webbook

ripol	1466.00		NIST Webbook
ripol	1481.00		NIST Webbook
tb	437.20	K	NIST Webbook
tc	644.50	K	Joback Method
tf	226.34	K	Joback Method
vc	0.309	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	173.65	J/mol×K	447.57	Joback Method
cpg	226.15	J/mol×K	611.68	Joback Method
cpg	216.79	J/mol×K	578.86	Joback Method
cpg	206.87	J/mol×K	546.03	Joback Method
cpg	196.39	J/mol×K	513.21	Joback Method
cpg	185.32	J/mol×K	480.39	Joback Method
cpg	234.98	J/mol×K	644.50	Joback Method
dvisc	0.0002345	Paxs	447.57	Joback Method
dvisc	0.0003970	Paxs	410.70	Joback Method
dvisc	0.0007454	Paxs	373.83	Joback Method
dvisc	0.0016066	Paxs	336.95	Joback Method
dvisc	0.0041819	Paxs	300.08	Joback Method
dvisc	0.0142316	Paxs	263.21	Joback Method
dvisc	0.0721805	Paxs	226.34	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C822673&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C822673&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

**cpg:** Ideal gas heat capacity

<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/19-906-5/2-Cyclohexen-1-ol.pdf>

Generated by Cheméo on 2024-04-26 08:04:04.915702438 +0000 UTC m=+16407893.836279752.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.