

# 2-Chlorobenzoic acid, heptyl ester

<b>Other names:</b>	Heptyl 2-chlorobenzoate Benzoic acid, 2-chloro, heptyl ester
<b>Inchi:</b>	InChI=1S/C14H19ClO2/c1-2-3-4-5-8-11-17-14(16)12-9-6-7-10-13(12)15/h6-7,9-10H,2-5,
<b>InchiKey:</b>	AZDUVJGZVLGFLA-UHFFFAOYSA-N
<b>Formula:</b>	C14H19ClO2
<b>SMILES:</b>	CCCCCCCOC(=O)c1cccc1Cl
<b>Mol. weight [g/mol]:</b>	254.75
<b>CAS:</b>	10276-86-5

## Physical Properties

Property code	Value	Unit	Source
gf	-76.07	kJ/mol	Joback Method
hf	-367.77	kJ/mol	Joback Method
hfus	32.65	kJ/mol	Joback Method
hvap	63.24	kJ/mol	Joback Method
log10ws	-4.91		Crippen Method
logp	4.467		Crippen Method
mvol	204.040	ml/mol	McGowan Method
pc	1998.33	kPa	Joback Method
rinpol	1818.00		NIST Webbook
rinpol	1861.90		NIST Webbook
rinpol	1861.90		NIST Webbook
rinpol	1841.00		NIST Webbook
rinpol	1836.00		NIST Webbook
rinpol	1826.00		NIST Webbook
rinpol	1816.00		NIST Webbook
rinpol	1826.00		NIST Webbook
rinpol	1831.00		NIST Webbook
ripol	2438.00		NIST Webbook
ripol	2446.00		NIST Webbook
ripol	2425.00		NIST Webbook
ripol	2459.00		NIST Webbook
ripol	2475.00		NIST Webbook
ripol	2452.00		NIST Webbook
ripol	2452.00		NIST Webbook
tb	665.10	K	Joback Method
tc	869.77	K	Joback Method

tf	388.56	K	Joback Method
vc	0.784	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	522.84	J/mol×K	665.10	Joback Method
cpg	589.90	J/mol×K	835.66	Joback Method
cpg	578.16	J/mol×K	801.55	Joback Method
cpg	565.61	J/mol×K	767.43	Joback Method
cpg	552.23	J/mol×K	733.32	Joback Method
cpg	537.98	J/mol×K	699.21	Joback Method
cpg	600.85	J/mol×K	869.77	Joback Method
dvisc	0.0001398	Paxs	665.10	Joback Method
dvisc	0.0001778	Paxs	619.01	Joback Method
dvisc	0.0002352	Paxs	572.92	Joback Method
dvisc	0.0003266	Paxs	526.83	Joback Method
dvisc	0.0004830	Paxs	480.74	Joback Method
dvisc	0.0007761	Paxs	434.65	Joback Method
dvisc	0.0013958	Paxs	388.56	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10276865&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10276865&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>r<sub>ipol</sub>:</b>	Polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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