

2-Phenylbutyryl chloride

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| Other names: | «alpha»-Phenylbutyryl chloride Benzeneacetyl chloride, «alpha»-ethyl- Butyryl chloride, 2-phenyl- |
| Inchi: | InChI=1S/C10H11ClO/c1-2-9(10(11)12)8-6-4-3-5-7-8/h3-7,9H,2H2,1H3 |
| InchiKey: | QGXMHCMPIAYMGT-UHFFFAOYSA-N |
| Formula: | C10H11ClO |
| SMILES: | CCC(C(=O)Cl)c1ccccc1 |
| Mol. weight [g/mol]: | 182.65 |
| CAS: | 36854-57-6 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | 2.44 | kJ/mol | Joback Method |
| hf | -146.80 | kJ/mol | Joback Method |
| hfus | 17.97 | kJ/mol | Joback Method |
| hvap | 50.87 | kJ/mol | Joback Method |
| log10ws | -3.00 | | Crippen Method |
| logp | 2.946 | | Crippen Method |
| mcvol | 141.810 | ml/mol | McGowan Method |
| pc | 3042.32 | kPa | Joback Method |
| rinpol | 1032.00 | | NIST Webbook |
| tb | 545.74 | K | Joback Method |
| tc | 770.42 | K | Joback Method |
| tf | 293.73 | K | Joback Method |
| vc | 0.536 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 303.48 | J/mol×K | 545.74 | Joback Method |
| cpg | 361.95 | J/mol×K | 732.97 | Joback Method |
| cpg | 351.90 | J/mol×K | 695.53 | Joback Method |
| cpg | 341.08 | J/mol×K | 658.08 | Joback Method |
| cpg | 329.43 | J/mol×K | 620.63 | Joback Method |

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|-------|-----------|---------|--------|---------------|
| cpg | 316.91 | J/molxK | 583.19 | Joback Method |
| cpg | 371.25 | J/molxK | 770.42 | Joback Method |
| dvisc | 0.0002491 | Paxs | 545.74 | Joback Method |
| dvisc | 0.0003250 | Paxs | 503.74 | Joback Method |
| dvisc | 0.0004450 | Paxs | 461.74 | Joback Method |
| dvisc | 0.0006490 | Paxs | 419.74 | Joback Method |
| dvisc | 0.0010293 | Paxs | 377.73 | Joback Method |
| dvisc | 0.0018320 | Paxs | 335.73 | Joback Method |
| dvisc | 0.0038452 | Paxs | 293.73 | Joback Method |

Pressure Dependent Properties

| Property code | Value | Unit | Pressure [kPa] | Source |
|---------------|--------|------|----------------|--------------|
| tbrp | 417.00 | K | 2.00 | NIST Webbook |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C36854576&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |

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|----------------|-----------------------------------|
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tbrp: | Boiling point at reduced pressure |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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