

Cyanogen

Other names:	(CN) ₂ CARBON NITRILE Carbon nitride Carbon nitride (C ₂ N ₂) Cyanogen (C ₂ N ₂) Cyanogen gas Cyanogene Dicyan Dicyanogen ETHANEDINITRILE NCCN Nitriloacetonitrile OXALONITRILE Oxalic acid dinitrile Oxalic nitrile Oxalyl cyanide Prussite Rcra waste number P031 UN 1026
Inchi:	InChI=1S/C ₂ N ₂ /c3-1-2-4
InchiKey:	JMANVNJQNLATNU-UHFFFAOYSA-N
Formula:	C ₂ N ₂
SMILES:	N#CC#N
Mol. weight [g/mol]:	52.03
CAS:	460-19-5

Physical Properties

Property code	Value	Unit	Source
af	0.2780		KDB
affp	674.70	kJ/mol	NIST Webbook
basg	621.70	kJ/mol	NIST Webbook
basg	645.80	kJ/mol	NIST Webbook
basg	642.00	kJ/mol	NIST Webbook
basg	622.80	kJ/mol	NIST Webbook
chg	-1096.00 ± 1.80	kJ/mol	NIST Webbook
chg	-1093.70 ± 0.84	kJ/mol	NIST Webbook
chg	-1051.90 ± 5.00	kJ/mol	NIST Webbook

dm	0.20	debye	KDB
gf	297.40	kJ/mol	KDB
hf	309.20	kJ/mol	KDB
hf	308.90 ± 1.80	kJ/mol	NIST Webbook
hf	263.00	kJ/mol	NIST Webbook
hf	306.70 ± 0.84	kJ/mol	NIST Webbook
hfus	3.95	kJ/mol	Joback Method
hvap	20.80	kJ/mol	NIST Webbook
ie	13.50	eV	NIST Webbook
ie	13.37 ± 0.01	eV	NIST Webbook
ie	13.50 ± 0.30	eV	NIST Webbook
ie	13.51	eV	NIST Webbook
ie	13.37 ± 0.01	eV	NIST Webbook
ie	13.36 ± 0.01	eV	NIST Webbook
log10ws	-0.39		Crippen Method
logp	0.034		Crippen Method
mcvol	41.800	ml/mol	McGowan Method
pc	5980.00	kPa	KDB
pt	75.10 ± 1.33	kPa	NIST Webbook
rinpola	273.00		NIST Webbook
rinpola	259.20		NIST Webbook
rinpola	259.20		NIST Webbook
sl	138.83	J/molxK	NIST Webbook
tb	252.00	K	NIST Webbook
tb	252.00	K	NIST Webbook
tb	252.00	K	KDB
tc	400.00	K	KDB
tf	245.20	K	KDB
tt	245.32 ± 0.02	K	NIST Webbook
vc	0.200	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	58.91	J/molxK	449.32	Joback Method
cpg	61.94	J/molxK	523.13	Joback Method
cpg	63.36	J/molxK	560.03	Joback Method
cpg	67.25	J/molxK	670.74	Joback Method
cpg	66.02	J/molxK	633.84	Joback Method
cpg	64.72	J/molxK	596.93	Joback Method
cpg	60.46	J/molxK	486.22	Joback Method

cpl	105.73	J/mol×K	255.00	NIST Webbook
hfust	8.11	kJ/mol	245.32	NIST Webbook
hfust	8.11	kJ/mol	245.30	NIST Webbook
hsubt	32.40	kJ/mol	219.00	NIST Webbook
hsubt	34.40	kJ/mol	223.50	NIST Webbook
hsubt	33.00	kJ/mol	220.50	NIST Webbook
hsubt	33.60	kJ/mol	203.50	NIST Webbook
hvapt	23.30 ± 0.20	kJ/mol	251.95	NIST Webbook
hvapt	23.50	kJ/mol	259.50	NIST Webbook
hvapt	23.90	kJ/mol	259.50	NIST Webbook
hvapt	24.50	kJ/mol	246.50	NIST Webbook
hvapt	23.33	kJ/mol	252.00	NIST Webbook
hvapt	23.33	kJ/mol	252.00	NIST Webbook
rho1	954.00	kg/m ³	252.00	KDB
sfust	33.05	J/mol×K	245.32	NIST Webbook
srf	0.02	N/m	217.00	KDB
svapt	92.58	J/mol×K	252.00	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.39745e+01
Coeff. B	-1.79426e+03
Coeff. C	-6.02280e+01
Temperature range (K), min.	191.32
Temperature range (K), max.	400.15

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
KDB:	https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1412
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C460195&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

af:	Acentric Factor
affp:	Proton affinity
basg:	Gas basicity
chg:	Standard gas enthalpy of combustion
cpg:	Ideal gas heat capacity
cpl:	Liquid phase heat capacity
dm:	Dipole Moment
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pt:	Triple Point Pressure
pvap:	Vapor pressure
rhol:	Liquid Density
rinpola:	Non-polar retention indices
sfust:	Entropy of fusion at a given temperature
sl:	Liquid phase molar entropy at standard conditions
srf:	Surface Tension
svapt:	Entropy of vaporization at a given temperature
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
tt:	Triple Point Temperature
vc:	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/20-049-5/Cyanogen.pdf>

Generated by Cheméo on 2024-02-22 21:38:45.315852931 +0000 UTC m=+10927174.236430253.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.