

# (p-Methoxyphenyl)phenylmethyl chloride

<b>Inchi:</b>	InChI=1S/C14H13ClO/c1-16-13-9-7-12(8-10-13)14(15)11-5-3-2-4-6-11/h2-10,14H,1H3
<b>InchiKey:</b>	FKRXYCHXGNKYPV-UHFFFAOYSA-N
<b>Formula:</b>	C14H13ClO
<b>SMILES:</b>	COc1ccc(C(Cl)c2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	232.71
<b>CAS:</b>	6731-11-9

## Physical Properties

Property code	Value	Unit	Source
gf	162.82	kJ/mol	Joback Method
hf	-23.94	kJ/mol	Joback Method
hfus	21.57	kJ/mol	Joback Method
hvap	58.38	kJ/mol	Joback Method
log10ws	-4.30		Crippen Method
logp	4.023		Crippen Method
mcvol	178.710	ml/mol	McGowan Method
pc	2600.43	kPa	Joback Method
tb	637.47	K	Joback Method
tc	882.99	K	Joback Method
tf	350.05	K	Joback Method
vc	0.664	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	425.52	J/mol×K	637.47	Joback Method
cpg	493.85	J/mol×K	842.07	Joback Method
cpg	482.45	J/mol×K	801.15	Joback Method
cpg	469.97	J/mol×K	760.23	Joback Method
cpg	456.36	J/mol×K	719.31	Joback Method
cpg	441.56	J/mol×K	678.39	Joback Method
cpg	504.24	J/mol×K	882.99	Joback Method
dvisc	0.0001275	Paxs	637.47	Joback Method
dvisc	0.0001640	Paxs	589.57	Joback Method

dvisc	0.0002206	Paxs	541.66	Joback Method
dvisc	0.0003143	Paxs	493.76	Joback Method
dvisc	0.0004830	Paxs	445.86	Joback Method
dvisc	0.0008234	Paxs	397.95	Joback Method
dvisc	0.0016242	Paxs	350.05	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6731119&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6731119&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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