

# 4-CHLOROPHENYL CYCLOPROPYL KETONE

|                      |                                                               |
|----------------------|---------------------------------------------------------------|
| Other names:         | Methanone, (4-chlorophenyl)cyclopropyl-                       |
| Inchi:               | InChI=1S/C10H9ClO/c11-9-5-3-8(4-6-9)10(12)7-1-2-7/h3-7H,1-2H2 |
| InchiKey:            | OPSFCTBBDIDFJM-UHFFFAOYSA-N                                   |
| Formula:             | C10H9ClO                                                      |
| SMILES:              | O=C(c1ccc(Cl)cc1)C1CC1                                        |
| Mol. weight [g/mol]: | 180.63                                                        |

## Physical Properties

| Property code | Value   | Unit   | Source             |
|---------------|---------|--------|--------------------|
| hf            | -75.42  | kJ/mol | Cheméo Relay (1.0) |
| hfus          | 19.24   | kJ/mol | Joback Method      |
| hvap          | 73.03   | kJ/mol | Cheméo Relay (1.0) |
| ie            | 9.24    | eV     | Cheméo Relay (1.0) |
| log10ws       | -3.65   |        | Cheméo Relay (1.0) |
| logp          | 2.933   |        | Crippen Method     |
| mcvol         | 130.950 | ml/mol | McGowan Method     |
| tb            | 545.24  | K      | Cheméo Relay (1.0) |
| tf            | 346.50  | K      | Cheméo Relay (1.0) |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 287.71    | J/mol×K | 557.90          | Joback Method |
| cpg           | 353.72    | J/mol×K | 795.94          | Joback Method |
| cpg           | 344.83    | J/mol×K | 756.26          | Joback Method |
| cpg           | 335.20    | J/mol×K | 716.59          | Joback Method |
| cpg           | 324.75    | J/mol×K | 676.92          | Joback Method |
| cpg           | 313.40    | J/mol×K | 637.25          | Joback Method |
| cpg           | 301.08    | J/mol×K | 597.57          | Joback Method |
| dvisc         | 0.0009432 | Pa×s    | 448.54          | Joback Method |
| dvisc         | 0.0005742 | Pa×s    | 557.90          | Joback Method |
| dvisc         | 0.0006620 | Pa×s    | 521.45          | Joback Method |
| dvisc         | 0.0007798 | Pa×s    | 485.00          | Joback Method |
| dvisc         | 0.0021340 | Pa×s    | 339.19          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0011801 | Pa×s | 412.09 | Joback Method |
| dvisc | 0.0015419 | Pa×s | 375.64 | Joback Method |

## Sources

Cheméo Relay (1.0, beta):

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C6640251&Units=SI>

Joback Method: [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method: [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Cheméo Relay (1.0):

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tf:</b>      | Normal melting (fusion) point                   |

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