

# (3-Iodophenyl) methanol, 3-methylbutyl ether

<b>Inchi:</b>	InChI=1S/C12H17IO/c1-10(2)6-7-14-9-11-4-3-5-12(13)8-11/h3-5,8,10H,6-7,9H2,1-2H3
<b>InchiKey:</b>	UUCSIXSLIAMCNH-UHFFFAOYSA-N
<b>Formula:</b>	C12H17IO
<b>SMILES:</b>	CC(C)CCOCc1cccc(I)c1
<b>Mol. weight [g/mol]:</b>	304.17

## Physical Properties

Property code	Value	Unit	Source
gf	103.62	kJ/mol	Joback Method
hf	-126.58	kJ/mol	Joback Method
hfus	22.56	kJ/mol	Joback Method
hvap	56.64	kJ/mol	Joback Method
log10ws	-4.44		Crippen Method
logp	3.854		Crippen Method
mcvol	187.870	ml/mol	McGowan Method
pc	2300.32	kPa	Joback Method
rinpol	1717.00		NIST Webbook
rinpol	1717.00		NIST Webbook
tb	620.74	K	Joback Method
tc	851.41	K	Joback Method
tf	329.23	K	Joback Method
vc	0.700	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	430.19	J/molxK	620.74	Joback Method
cpg	498.57	J/molxK	812.97	Joback Method
cpg	486.72	J/molxK	774.52	Joback Method
cpg	474.00	J/molxK	736.08	Joback Method
cpg	460.36	J/molxK	697.63	Joback Method
cpg	445.77	J/molxK	659.19	Joback Method
cpg	509.59	J/molxK	851.41	Joback Method
dvisc	0.0001500	Paxs	620.74	Joback Method

dvisc	0.0001956	Paxs	572.15	Joback Method
dvisc	0.0002678	Paxs	523.57	Joback Method
dvisc	0.0003911	Paxs	474.99	Joback Method
dvisc	0.0006225	Paxs	426.40	Joback Method
dvisc	0.0011168	Paxs	377.81	Joback Method
dvisc	0.0023806	Paxs	329.23	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U374577&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U374577&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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