

3-Hexene, 2,2,5,5-tetramethyl-, (Z)-

Other names:	(Z)-2,2,5,5-Tetramethyl-3-hexene cis-Di-tert-butylethylene cis-1,2-Di-tert-butylethylene cis-2,2,5,5-Tetramethyl-3-hexene (Z)-(t-C4H9)CH=CH(t-C4H9) (Z)-1,2-Di-tert-butylethylene
Inchi:	InChI=1S/C10H20/c1-9(2,3)7-8-10(4,5)6/h7-8H,1-6H3/b8-7-
InchiKey:	UYWLQEPMPVDASH-FPLPWBNLSA-N
Formula:	C10H20
SMILES:	CC(C)(C)C=CC(C)(C)C
Mol. weight [g/mol]:	140.27
CAS:	692-47-7

Physical Properties

Property code	Value	Unit	Source
chl	-6629.80 ± 1.70	kJ/mol	NIST Webbook
gf	119.22	kJ/mol	Joback Method
hf	-127.00	kJ/mol	NIST Webbook
hfus	7.03	kJ/mol	Joback Method
hvap	35.22	kJ/mol	Joback Method
ie	8.68 ± 0.02	eV	NIST Webbook
ie	8.95	eV	NIST Webbook
ie	8.70 ± 0.01	eV	NIST Webbook
log10ws	-3.38		Crippen Method
logp	3.635		Crippen Method
mcvol	147.460	ml/mol	McGowan Method
pc	2293.71	kPa	Joback Method
tb	417.00 ± 4.00	K	NIST Webbook
tb	416.00 ± 1.50	K	NIST Webbook
tb	416.00 ± 3.00	K	NIST Webbook
tc	618.91	K	Joback Method
tf	202.22	K	Joback Method
vc	0.553	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	303.83	J/mol×K	425.90	Joback Method
cpg	383.91	J/mol×K	586.75	Joback Method
cpg	369.87	J/mol×K	554.58	Joback Method
cpg	354.91	J/mol×K	522.41	Joback Method
cpg	338.96	J/mol×K	490.24	Joback Method
cpg	321.95	J/mol×K	458.07	Joback Method
cpg	397.07	J/mol×K	618.91	Joback Method
dvisc	0.0002185	Pa×s	425.90	Joback Method
dvisc	0.0003211	Pa×s	388.62	Joback Method
dvisc	0.0005120	Pa×s	351.34	Joback Method
dvisc	0.0009122	Pa×s	314.06	Joback Method
dvisc	0.0018988	Pa×s	276.78	Joback Method
dvisc	0.0049656	Pa×s	239.50	Joback Method
dvisc	0.0185097	Pa×s	202.22	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C692477&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy

log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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