

# 1,2-Heptanediol

Inchi:	InChI=1S/C7H16O2/c1-2-3-4-5-7(9)6-8/h7-9H,2-6H2,1H3
InchiKey:	GCXZDAKFJKCPGK-UHFFFAOYSA-N
Formula:	C7H16O2
SMILES:	CCCCC(O)CO
Mol. weight [g/mol]:	132.20
CAS:	3710-31-4

## Physical Properties

Property code	Value	Unit	Source
gf	-268.02	kJ/mol	Joback Method
hf	-497.55	kJ/mol	Joback Method
hfus	18.54	kJ/mol	Joback Method
hvap	64.15	kJ/mol	Joback Method
log10ws	-1.39		Crippen Method
logp	0.920		Crippen Method
mcvol	121.230	ml/mol	McGowan Method
pc	3468.36	kPa	Joback Method
rinpol	1098.00		NIST Webbook
rinpol	1065.00		NIST Webbook
rinpol	1065.00		NIST Webbook
ripol	1470.00		NIST Webbook
ripol	1470.00		NIST Webbook
tb	543.48	K	Joback Method
tc	703.08	K	Joback Method
tf	275.29	K	Joback Method
vc	0.460	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	299.37	J/molxK	543.48	Joback Method
cpg	308.91	J/molxK	570.08	Joback Method
cpg	318.08	J/molxK	596.68	Joback Method
cpg	326.88	J/molxK	623.28	Joback Method

cpg	335.33	J/mol×K	649.88	Joback Method
cpg	343.43	J/mol×K	676.48	Joback Method
cpg	351.20	J/mol×K	703.08	Joback Method
dvisc	0.1642467	Paxs	275.29	Joback Method
dvisc	0.0172033	Paxs	319.99	Joback Method
dvisc	0.0031328	Paxs	364.69	Joback Method
dvisc	0.0008275	Paxs	409.38	Joback Method
dvisc	0.0002841	Paxs	454.08	Joback Method
dvisc	0.0001181	Paxs	498.78	Joback Method
dvisc	0.0000567	Paxs	543.48	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.68850e+01
Coeff. B	-5.18952e+03
Coeff. C	-8.32500e+01
Temperature range (K), min.	395.92
Temperature range (K), max.	531.65

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3710314&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3710314&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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