

# Adipic acid, 3-oxobut-2-yl tridecyl ester

<b>Inchi:</b>	InChI=1S/C23H42O5/c1-4-5-6-7-8-9-10-11-12-13-16-19-27-22(25)17-14-15-18-23(26)28
<b>InchiKey:</b>	JSAAKZPEPQMEII-UHFFFAOYSA-N
<b>Formula:</b>	C23H42O5
<b>SMILES:</b>	CCCCCCCCCCCCOC(=O)CCCC(=O)OC(C)C(C)=O
<b>Mol. weight [g/mol]:</b>	398.58

## Physical Properties

Property code	Value	Unit	Source
gf	-456.42	kJ/mol	Joback Method
hf	-1125.51	kJ/mol	Joback Method
hfus	58.98	kJ/mol	Joback Method
hvap	91.46	kJ/mol	Joback Method
log10ws	-6.57		Crippen Method
logp	5.922		Crippen Method
mcvol	351.380	ml/mol	McGowan Method
pc	939.22	kPa	Joback Method
rinpol	2724.00		NIST Webbook
tb	931.65	K	Joback Method
tc	1141.56	K	Joback Method
tf	528.22	K	Joback Method
vc	1.371	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1176.67	J/molxK	931.65	Joback Method
cpg	1194.68	J/molxK	966.64	Joback Method
cpg	1211.25	J/molxK	1001.62	Joback Method
cpg	1226.41	J/molxK	1036.61	Joback Method
cpg	1240.19	J/molxK	1071.59	Joback Method
cpg	1252.63	J/molxK	1106.58	Joback Method
cpg	1263.75	J/molxK	1141.56	Joback Method
dvisc	0.0005155	Paxs	528.22	Joback Method
dvisc	0.0002441	Paxs	595.46	Joback Method

dvisc	0.0001345	Paxs	662.70	Joback Method
dvisc	0.0000828	Paxs	729.93	Joback Method
dvisc	0.0000552	Paxs	797.17	Joback Method
dvisc	0.0000393	Paxs	864.41	Joback Method
dvisc	0.0000293	Paxs	931.65	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353757&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353757&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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