

# Benzene, 1-fluoro-4-(trifluoromethyl)-

<b>Other names:</b>	1-Fluoro-4-(trifluoromethyl)benzene 4-Fluorobenzotrifluoride NSC 88289 Toluene, p, «alpha», «alpha», «alpha»-tetrafluoro- Toluene, p, «alpha», «alpha», «alpha»-tetrafluoro- p, «alpha», «alpha», «alpha»-Tetrafluorotoluene p, «alpha», «alpha», «alpha»-Tetrafluorotoluene p-(Trifluoromethyl)fluorobenzene p-Fluorobenzotrifluoride p-Fluorotrifluoromethylbenzene «alpha», «alpha», «alpha», 4-tetrafluorotoluene «alpha», «alpha», «alpha», p-Tetrafluorotoluene «alpha», «alpha», «alpha», 4-tetrafluorotoluene «alpha», «alpha», «alpha», p-Tetrafluorotoluene
<b>Inchi:</b>	InChI=1S/C7H4F4/c8-6-3-1-5(2-4-6)7(9,10)11/h1-4H
<b>InchiKey:</b>	UNNNAIWPDLRVRN-UHFFFAOYSA-N
<b>Formula:</b>	C7H4F4
<b>SMILES:</b>	Fc1ccc(C(F)(F)F)cc1
<b>Mol. weight [g/mol]:</b>	164.10
<b>CAS:</b>	402-44-8

## Physical Properties

Property code	Value	Unit	Source
gf	-665.56	kJ/mol	Joback Method
hf	-755.94	kJ/mol	Joback Method
hfus	12.44	kJ/mol	Joback Method
hvap	29.55	kJ/mol	Joback Method
ie	9.98	eV	NIST Webbook
log10ws	-2.91		Crippen Method
logp	2.845		Crippen Method
mcvol	92.810	ml/mol	McGowan Method
pc	3322.01	kPa	Joback Method
tb	376.70	K	NIST Webbook
tb	375.90	K	NIST Webbook
tc	566.36	K	Joback Method
tf	212.37	K	Joback Method
vc	0.381	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	219.11	J/mol×K	536.14	Joback Method
cpg	174.38	J/mol×K	385.07	Joback Method
cpg	184.55	J/mol×K	415.28	Joback Method
cpg	194.08	J/mol×K	445.50	Joback Method
cpg	203.00	J/mol×K	475.71	Joback Method
cpg	211.34	J/mol×K	505.93	Joback Method
cpg	226.36	J/mol×K	566.36	Joback Method
hvapt	35.80	kJ/mol	333.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.64671e+01
Coeff. B	-4.03068e+03
Coeff. C	-2.35950e+01
Temperature range (K), min.	272.72
Temperature range (K), max.	384.91

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C402448&Units=SI>

**The Yaws Handbook of Vapor Pressure:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvac:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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