

# Diethyl azodicarboxylate

<b>Other names:</b>	Ethyl azodicarboxylate Diazenedicarboxylic acid, diethyl ester Diethoxycarbonyldiazene Diethyl azodiformate Diethyl diazodicarboxylate Formic acid, azodi-, diethyl ester
<b>Inchi:</b>	InChI=1S/C6H10N2O4/c1-3-11-5(9)7-8-6(10)12-4-2/h3-4H2,1-2H3
<b>InchiKey:</b>	FAMRKDQNMBBFBR-UHFFFAOYSA-N
<b>Formula:</b>	C6H10N2O4
<b>SMILES:</b>	CCOC(=O)N=NC(=O)OCC
<b>Mol. weight [g/mol]:</b>	174.15
<b>CAS:</b>	1972-28-7

## Physical Properties

Property code	Value	Unit	Source
hf	-609.55	kJ/mol	Joback Method
hvap	53.93	kJ/mol	Joback Method
log10ws	-1.27		Crippen Method
logp	1.752		Crippen Method
mcvol	125.940	ml/mol	McGowan Method
pc	2592.49	kPa	Joback Method
tb	638.46	K	Joback Method
tc	849.78	K	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1972287&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1972287&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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