

# 4-Amino-1-butanol

<b>Other names:</b>	4-hydroxybutylamine H <sub>2</sub> NCH <sub>2</sub> (CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> OH 4-Aminobutanol-1 1-Butanol, 4-amino- 4-Aminobutanol NH <sub>2</sub> (CH <sub>2</sub> ) <sub>4</sub> OH
<b>Inchi:</b>	InChI=1S/C4H11NO/c5-3-1-2-4-6/h6H,1-5H2
<b>InchiKey:</b>	BLFRQYKZFKYQLO-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>4</sub> H <sub>11</sub> NO
<b>SMILES:</b>	NCCCCO
<b>Mol. weight [g/mol]:</b>	89.14
<b>CAS:</b>	13325-10-5

## Physical Properties

Property code	Value	Unit	Source
affp	984.50	kJ/mol	NIST Webbook
basg	932.10	kJ/mol	NIST Webbook
gf	-87.57	kJ/mol	Joback Method
hf	-244.33	kJ/mol	Joback Method
hfus	15.40	kJ/mol	Joback Method
hvap	51.82	kJ/mol	Joback Method
log10ws	-0.19		Crippen Method
logp	-0.282		Crippen Method
mcvol	83.070	ml/mol	McGowan Method
pc	4736.62	kPa	Joback Method
rinpol	998.00		NIST Webbook
rinpol	904.00		NIST Webbook
rinpol	904.00		NIST Webbook
ripol	1687.00		NIST Webbook
tb	479.00	K	NIST Webbook
tb	476.00 ± 6.00	K	NIST Webbook
tb	479.20	K	NIST Webbook
tc	631.39	K	Joback Method
tf	278.92	K	Joback Method
vc	0.307	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	179.46	J/molxK	455.63	Joback Method
cpg	187.25	J/molxK	484.92	Joback Method
cpg	194.72	J/molxK	514.22	Joback Method
cpg	201.89	J/molxK	543.51	Joback Method
cpg	208.76	J/molxK	572.80	Joback Method
cpg	215.33	J/molxK	602.09	Joback Method
cpg	221.63	J/molxK	631.39	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	398.20	K	4.50	NIST Webbook
tbrp	398.00	K	4.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13325105&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13325105&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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