

2-Ethylbutyric acid, 4-fluoro-2-methoxyphenyl ester

Inchi:	InChI=1S/C13H17FO3/c1-4-9(5-2)13(15)17-11-7-6-10(14)8-12(11)16-3/h6-9H,4-5H2,1-3
InchiKey:	QZGLLTJQEQQJH-UHFFFAOYSA-N
Formula:	C13H17FO3
SMILES:	CCC(CC)C(=O)Oc1ccc(F)cc1OC
Mol. weight [g/mol]:	240.27

Physical Properties

Property code	Value	Unit	Source
gf	-384.44	kJ/mol	Joback Method
hf	-676.47	kJ/mol	Joback Method
hfus	26.22	kJ/mol	Joback Method
hvap	58.49	kJ/mol	Joback Method
log10ws	-3.67		Crippen Method
logp	3.176		Crippen Method
mvol	185.350	ml/mol	McGowan Method
pc	2117.78	kPa	Joback Method
rinpol	1559.00		NIST Webbook
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tb	631.02	K	Joback Method
tc	828.98	K	Joback Method
tf	367.71	K	Joback Method
vc	0.710	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	477.44	J/molxK	631.02	Joback Method
cpg	492.16	J/molxK	664.01	Joback Method
cpg	506.11	J/molxK	697.01	Joback Method
cpg	519.31	J/molxK	730.00	Joback Method
cpg	531.75	J/molxK	762.99	Joback Method
cpg	543.44	J/molxK	795.98	Joback Method
cpg	554.37	J/molxK	828.98	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U370922&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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