

# Ethyl cyclopropanecarboxylate

<b>Other names:</b>	Cyclopropanecarboxylic acid ethyl ester Ethyl cyclopropylcarboxylate
<b>Inchi:</b>	InChI=1S/C6H10O2/c1-2-8-6(7)5-3-4-5/h5H,2-4H2,1H3
<b>InchiKey:</b>	LDDOSDVZPSGLFZ-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O2
<b>SMILES:</b>	CCOC(=O)C1CC1
<b>Mol. weight [g/mol]:</b>	114.14
<b>CAS:</b>	4606-07-9

## Physical Properties

Property code	Value	Unit	Source
gf	-173.53	kJ/mol	Joback Method
hf	-339.17	kJ/mol	Joback Method
hfus	12.22	kJ/mol	Joback Method
hvap	38.02	kJ/mol	Joback Method
log10ws	-0.85		Crippen Method
logp	0.959		Crippen Method
mvol	91.980	ml/mol	McGowan Method
pc	3819.82	kPa	Joback Method
tb	404.20	K	NIST Webbook
tc	612.36	K	Joback Method
tf	247.48	K	Joback Method
vc	0.352	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	242.55	J/molxK	612.36	Joback Method
cpg	233.99	J/molxK	580.25	Joback Method
cpg	224.95	J/molxK	548.14	Joback Method
cpg	215.43	J/molxK	516.03	Joback Method
cpg	205.41	J/molxK	483.93	Joback Method
cpg	194.86	J/molxK	451.82	Joback Method
cpg	183.77	J/molxK	419.71	Joback Method

cpl	213.00	J/mol×K	298.15	NIST Webbook
dvisc	0.0004994	Paxs	391.00	Joback Method
dvisc	0.0005788	Paxs	362.30	Joback Method
dvisc	0.0006880	Paxs	333.60	Joback Method
dvisc	0.0008449	Paxs	304.89	Joback Method
dvisc	0.0010828	Paxs	276.19	Joback Method
dvisc	0.0014699	Paxs	247.48	Joback Method
dvisc	0.0004397	Paxs	419.71	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4606079&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4606079&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/20-650-7/Ethyl-cyclopropanecarboxylate.pdf>

Generated by Cheméo on 2024-04-20 03:45:29.904926148 +0000 UTC m=+15873978.825503460.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.