

valeric acid, 3-methylbut-2-enyl ester

Inchi: InChI=1S/C10H18O2/c1-4-5-6-10(11)12-8-7-9(2)3/h7H,4-6,8H2,1-3H3
InchiKey: PJMJLIYPHMEZSG-UHFFFAOYSA-N
Formula: C10H18O2
SMILES: CCCCC(=O)OCC=C(C)C
Mol. weight [g/mol]: 170.25

Physical Properties

Property code	Value	Unit	Source
af	0.5126		Cheméo Relay (1.0)
hf	-478.12	kJ/mol	Cheméo Relay (1.0)
hfus	23.34	kJ/mol	Joback Method
hvap	56.91	kJ/mol	Cheméo Relay (1.0)
ie	8.71	eV	Cheméo Relay (1.0)
log10ws	-2.96		Cheméo Relay (1.0)
logp	2.686		Crippen Method
mcvol	154.900	ml/mol	McGowan Method
pc	2322.54	kPa	Joback Method
rinpol	1195.60		NIST Webbook
tb	470.38	K	Cheméo Relay (1.0)
tc	645.60	K	Cheméo Relay (1.0)
tf	210.97	K	Cheméo Relay (1.0)
vc	0.570	m3/kmol	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.59	J/mol×K	508.53	Joback Method
cpg	364.62	J/mol×K	539.01	Joback Method
cpg	378.03	J/mol×K	569.50	Joback Method
cpg	390.86	J/mol×K	599.98	Joback Method
cpg	403.11	J/mol×K	630.47	Joback Method
cpg	414.81	J/mol×K	660.95	Joback Method
cpg	425.96	J/mol×K	691.44	Joback Method

Sources

Cheméo Relay (1.0, beta):

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=U292485&Units=SI>

Joback Method: https://en.wikipedia.org/wiki/Joback_method

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Cheméo Relay (1.0):

Legend

af:	Acentric Factor
cpg:	Ideal gas heat capacity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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