

# Carbonic acid, 2-methoxyethyl 3,4-dichlorophenyl ester

<b>Inchi:</b>	InChI=1S/C10H10Cl2O4/c1-14-4-5-15-10(13)16-7-2-3-8(11)9(12)6-7/h2-3,6H,4-5H2,1H3
<b>InchiKey:</b>	NFKKCZSBYDIYRC-UHFFFAOYSA-N
<b>Formula:</b>	C10H10Cl2O4
<b>SMILES:</b>	COCCOC(=O)Oc1ccc(Cl)c(Cl)c1
<b>Mol. weight [g/mol]:</b>	265.09

## Physical Properties

Property code	Value	Unit	Source
gf	-341.31	kJ/mol	Joback Method
hf	-576.86	kJ/mol	Joback Method
hfus	28.48	kJ/mol	Joback Method
hvap	64.20	kJ/mol	Joback Method
log10ws	-3.14		Crippen Method
logp	3.155		Crippen Method
mcvol	171.660	ml/mol	McGowan Method
pc	2659.77	kPa	Joback Method
rinqol	1819.00		NIST Webbook
tb	660.83	K	Joback Method
tc	878.25	K	Joback Method
tf	430.38	K	Joback Method
vc	0.645	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	395.62	J/molxK	660.83	Joback Method
cpg	406.78	J/molxK	697.07	Joback Method
cpg	417.26	J/molxK	733.30	Joback Method
cpg	427.05	J/molxK	769.54	Joback Method
cpg	436.12	J/molxK	805.77	Joback Method
cpg	444.46	J/molxK	842.01	Joback Method
cpg	452.04	J/molxK	878.25	Joback Method
dvisc	0.0006912	Paxs	430.38	Joback Method
dvisc	0.0004601	Paxs	468.79	Joback Method

dvisc	0.0003257	Paxs	507.20	Joback Method
dvisc	0.0002421	Paxs	545.61	Joback Method
dvisc	0.0001871	Paxs	584.01	Joback Method
dvisc	0.0001492	Paxs	622.42	Joback Method
dvisc	0.0001222	Paxs	660.83	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357875&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357875&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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