

# 3-Phenoxypropionic acid

<b>Other names:</b>	3-Phenoxypropanoic acid «beta»-Phenoxypropionic acid Propanoic acid, 3-phenoxy-
<b>Inchi:</b>	InChI=1S/C9H10O3/c10-9(11)6-7-12-8-4-2-1-3-5-8/h1-5H,6-7H2,(H,10,11)
<b>InchiKey:</b>	BUSOTUQRURCMCM-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O3
<b>SMILES:</b>	O=C(O)CCOc1ccccc1
<b>Mol. weight [g/mol]:</b>	166.17
<b>CAS:</b>	7170-38-9

## Physical Properties

Property code	Value	Unit	Source
gf	-233.43	kJ/mol	Joback Method
hf	-389.59	kJ/mol	Joback Method
hfus	19.98	kJ/mol	Joback Method
hvap	63.74	kJ/mol	Joback Method
log10ws	-1.53		Crippen Method
logp	1.540		Crippen Method
mcvol	127.220	ml/mol	McGowan Method
pc	3843.54	kPa	Joback Method
tb	513.20	K	NIST Webbook
tc	801.75	K	Joback Method
tf	350.59	K	Joback Method
vc	0.474	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	304.10	J/molxK	600.47	Joback Method
cpg	349.97	J/molxK	768.21	Joback Method
cpg	341.95	J/molxK	734.66	Joback Method
cpg	333.37	J/molxK	701.11	Joback Method
cpg	324.21	J/molxK	667.56	Joback Method
cpg	314.46	J/molxK	634.02	Joback Method

cpg	357.45	J/mol×K	801.75	Joback Method
dvisc	0.0000813	Paxs	600.47	Joback Method
dvisc	0.0001222	Paxs	558.82	Joback Method
dvisc	0.0001960	Paxs	517.18	Joback Method
dvisc	0.0003415	Paxs	475.53	Joback Method
dvisc	0.0006620	Paxs	433.88	Joback Method
dvisc	0.0014770	Paxs	392.24	Joback Method
dvisc	0.0039872	Paxs	350.59	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7170389&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7170389&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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