

# 4,5-Isoxazoledicarboxylic acid, 4,5-dihydro-3-phenyl-, dimethyl ester, cis-

Other names:	2-Isoxazoline-4,5-dicarboxylic acid, 3-phenyl-, dimethyl ester, cis-
Inchi:	InChI=1S/C13H13NO5/c1-17-12(15)9-10(8-6-4-3-5-7-8)14-19-11(9)13(16)18-2/h3-7,9,11
InchiKey:	RZUSIVILFUXVKH-KOLCDFICSA-N
Formula:	C13H13NO5
SMILES:	COC(=O)C1ON=C(c2ccccc2)C1C(=O)OC
Mol. weight [g/mol]:	263.25
CAS:	17669-30-6

## Physical Properties

Property code	Value	Unit	Source
chs	-6109.00	kJ/mol	NIST Webbook
gf	-217.02	kJ/mol	Joback Method
hf	-539.30	kJ/mol	Joback Method
hfs	-864.80	kJ/mol	NIST Webbook
hfus	38.00	kJ/mol	Joback Method
hvap	76.74	kJ/mol	Joback Method
log10ws	-1.20		Crippen Method
logp	0.752		Crippen Method
mcvol	185.840	ml/mol	McGowan Method
pc	2820.33	kPa	Joback Method
tb	771.50	K	Joback Method
tc	1011.41	K	Joback Method
tf	525.06	K	Joback Method
vc	0.700	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	553.79	J/molxK	771.50	Joback Method
cpg	568.16	J/molxK	811.49	Joback Method
cpg	581.10	J/molxK	851.47	Joback Method
cpg	592.58	J/molxK	891.46	Joback Method
cpg	602.59	J/molxK	931.44	Joback Method
cpg	611.12	J/molxK	971.43	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17669306&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17669306&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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