

# Desmethoxyencecalin

<b>Other names:</b>	2,2-Dimethyl-6-acetyl-3-chromene Demethoxyencecaline
<b>Inchi:</b>	InChI=1S/C13H14O2/c1-9(14)10-4-5-12-11(8-10)6-7-13(2,3)15-12/h4-8H,1-3H3
<b>InchiKey:</b>	ZAJTXVHECZCXLH-UHFFFAOYSA-N
<b>Formula:</b>	C13H14O2
<b>SMILES:</b>	CC(=O)c1ccc2c(c1)C=CC(C)(C)O2
<b>Mol. weight [g/mol]:</b>	202.25

## Physical Properties

Property code	Value	Unit	Source
gf	9.81	kJ/mol	Joback Method
hf	-202.98	kJ/mol	Joback Method
hfus	23.23	kJ/mol	Joback Method
hvap	58.61	kJ/mol	Joback Method
log10ws	-3.92		Crippen Method
logp	3.073		Crippen Method
mcvol	162.550	ml/mol	McGowan Method
pc	2844.44	kPa	Joback Method
ripol	1644.00		NIST Webbook
ripol	1646.00		NIST Webbook
ripol	1647.00		NIST Webbook
ripol	1657.00		NIST Webbook
ripol	1618.00		NIST Webbook
ripol	1657.00		NIST Webbook
ripol	1647.00		NIST Webbook
ripol	1646.00		NIST Webbook
ripol	2401.00		NIST Webbook
ripol	2401.00		NIST Webbook
ripol	2418.00		NIST Webbook
ripol	2449.00		NIST Webbook
tb	624.71	K	Joback Method
tc	862.61	K	Joback Method
tf	403.31	K	Joback Method
vc	0.616	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	410.23	J/mol×K	624.71	Joback Method
cpg	425.15	J/mol×K	664.36	Joback Method
cpg	439.13	J/mol×K	704.01	Joback Method
cpg	452.36	J/mol×K	743.66	Joback Method
cpg	464.99	J/mol×K	783.31	Joback Method
cpg	477.22	J/mol×K	822.96	Joback Method
cpg	489.22	J/mol×K	862.61	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R37656&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R37656&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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