

# 2-Propanol, 1-chloro-3-(1-naphthalenyloxy)-

**Other names:**

1-Chloro-3-(1-naphthoxy)-2-propanol  
1-(Naphth-1-yloxy)-3-chlor-propan-2-ol  
1-(Naphth-1-yloxy)-3-chloro-propan-2-ol  
1-Chloro-3-(1-naphthyloxy)-2-propanol  
2-Propanol, 1-chloro-3-(1-naphthyloxy)-  
1-chloro-3-(1-naphthyloxy)propan-2-ol

**Inchi:** InChI=1S/C13H13ClO2/c14-8-11(15)9-16-13-7-3-5-10-4-1-2-6-12(10)13/h1-7,11,15H,8-9**InchiKey:** ZVVJOJMCXYDDEW-UHFFFAOYSA-N**Formula:** C13H13ClO2**SMILES:** OC(CCl)COc1cccc2cccc12**Mol. weight [g/mol]:** 236.69**CAS:** 20133-93-1

## Physical Properties

Property code	Value	Unit	Source
gf	11.82	kJ/mol	Joback Method
hf	-200.99	kJ/mol	Joback Method
hfus	26.05	kJ/mol	Joback Method
hvap	72.20	kJ/mol	Joback Method
log10ws	-3.76		Crippen Method
logp	2.818		Crippen Method
mcvol	174.790	ml/mol	McGowan Method
pc	2893.62	kPa	Joback Method
tb	699.07	K	Joback Method
tc	913.27	K	Joback Method
tf	405.88	K	Joback Method
vc	0.657	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	446.20	J/molxK	699.07	Joback Method
cpg	497.40	J/molxK	877.57	Joback Method
cpg	488.55	J/molxK	841.87	Joback Method

cpg	479.05	J/molxK	806.17	Joback Method
cpg	468.86	J/molxK	770.47	Joback Method
cpg	457.93	J/molxK	734.77	Joback Method
cpg	505.68	J/molxK	913.27	Joback Method
dvisc	0.0000611	Paxs	699.07	Joback Method
dvisc	0.0000868	Paxs	650.21	Joback Method
dvisc	0.0001306	Paxs	601.34	Joback Method
dvisc	0.0002112	Paxs	552.48	Joback Method
dvisc	0.0003751	Paxs	503.61	Joback Method
dvisc	0.0007537	Paxs	454.75	Joback Method
dvisc	0.0017913	Paxs	405.88	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20133931&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20133931&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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