

# 3,4-Methylenedioxybenzyl acetate

<b>Other names:</b>	Piperonyl acetate 1,3-Benzodioxole-5-methanol, acetate Acetic acid, (3,4-methylenedioxy)benzyl ester Heliotropyl acetate Piperonyl alcohol, acetate 1,3-Benzodioxol, 5-(acetoxymethyl) Piperonal acetate
<b>Inchi:</b>	InChI=1S/C10H10O4/c1-7(11)12-5-8-2-3-9-10(4-8)14-6-13-9/h2-4H,5-6H2,1H3
<b>InchiKey:</b>	PFWYHTORQZAGCA-UHFFFAOYSA-N
<b>Formula:</b>	C10H10O4
<b>SMILES:</b>	CC(=O)OCc1ccc2c(c1)OCO2
<b>Mol. weight [g/mol]:</b>	194.18
<b>CAS:</b>	326-61-4

## Physical Properties

Property code	Value	Unit	Source
gf	-211.23	kJ/mol	Joback Method
hf	-451.80	kJ/mol	Joback Method
hfus	30.73	kJ/mol	Joback Method
hvap	59.85	kJ/mol	Joback Method
log10ws	-2.26		Crippen Method
logp	1.478		Crippen Method
mcvol	136.320	ml/mol	McGowan Method
pc	3517.91	kPa	Joback Method
rinpol	1485.00		NIST Webbook
rinpol	1514.00		NIST Webbook
rinpol	1514.00		NIST Webbook
rinpol	1530.00		NIST Webbook
rinpol	1485.00		NIST Webbook
ripol	2325.00		NIST Webbook
ripol	2344.00		NIST Webbook
ripol	2325.00		NIST Webbook
ripol	2344.00		NIST Webbook
tb	606.44	K	Joback Method
tc	832.75	K	Joback Method
tf	401.40	K	Joback Method
vc	0.511	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	338.90	J/molxK	606.44	Joback Method
cpg	350.92	J/molxK	644.16	Joback Method
cpg	362.13	J/molxK	681.88	Joback Method
cpg	372.58	J/molxK	719.59	Joback Method
cpg	382.30	J/molxK	757.31	Joback Method
cpg	391.33	J/molxK	795.03	Joback Method
cpg	399.73	J/molxK	832.75	Joback Method
dvisc	0.0017744	Paxs	401.40	Joback Method
dvisc	0.0012636	Paxs	435.57	Joback Method
dvisc	0.0009454	Paxs	469.75	Joback Method
dvisc	0.0007358	Paxs	503.92	Joback Method
dvisc	0.0005911	Paxs	538.09	Joback Method
dvisc	0.0004875	Paxs	572.27	Joback Method
dvisc	0.0004109	Paxs	606.44	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C326614&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C326614&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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