

# 1H-Pyrrole-2,5-dione, 3-ethyl-4-methyl-

<b>Other names:</b>	Maleimide, 2-ethyl-3-methyl- Ethylmethylmaleimide Methylethylmaleimide 2-Ethyl-3-methylmaleimide 3-Ethyl-4-methyl-1H-pyrrole-2,5-dione 3-Ethyl-4-methyl-2,5-pyrrolidine-dione 3-ethyl-4-methylmaleimide
<b>Inchi:</b>	InChI=1S/C7H9NO2/c1-3-5-4(2)6(9)8-7(5)10/h3H2,1-2H3,(H,8,9,10)
<b>InchiKey:</b>	CUBICSJJYOPOIA-UHFFFAOYSA-N
<b>Formula:</b>	C7H9NO2
<b>SMILES:</b>	CCC1=C(C)C(=O)NC1=O
<b>Mol. weight [g/mol]:</b>	139.15
<b>CAS:</b>	20189-42-8

## Physical Properties

Property code	Value	Unit	Source
gf	-94.45	kJ/mol	Joback Method
hf	-309.74	kJ/mol	Joback Method
hfus	15.80	kJ/mol	Joback Method
hvap	48.61	kJ/mol	Joback Method
log10ws	-1.24		Crippen Method
logp	0.369		Crippen Method
mcvol	107.450	ml/mol	McGowan Method
pc	4026.13	kPa	Joback Method
rinpol	1238.00		NIST Webbook
rinpol	1239.00		NIST Webbook
rinpol	1265.00		NIST Webbook
rinpol	1238.00		NIST Webbook
rinpol	1280.60		NIST Webbook
rinpol	1265.00		NIST Webbook
rinpol	1234.80		NIST Webbook
rinpol	1192.00		NIST Webbook
ripol	2244.00		NIST Webbook
ripol	2244.00		NIST Webbook
ripol	2260.00		NIST Webbook
ripol	2260.00		NIST Webbook
ripol	2197.00		NIST Webbook

tb	572.82	K	Joback Method
tc	814.55	K	Joback Method
tf	451.06	K	Joback Method
vc	0.406	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	249.38	J/mol×K	572.82	Joback Method
cpg	261.98	J/mol×K	613.11	Joback Method
cpg	274.10	J/mol×K	653.40	Joback Method
cpg	285.68	J/mol×K	693.68	Joback Method
cpg	296.65	J/mol×K	733.97	Joback Method
cpg	306.95	J/mol×K	774.26	Joback Method
cpg	316.51	J/mol×K	814.55	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20189428&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20189428&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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