

TATP

| | |
|-----------------------------|--|
| Other names: | 1,2,4,5,7,8-Hexaoxacyclononane, 3,3,6,6,9,9-hexamethyl 3,3,6,6,9,9-Hexamethyl-1,2,4,5,7,8-hexoxanane Acetone cyclic triperoxide Acetone peroxide trimer TATP (triacetone triperoxide) TCAP Triacetontriperoxide Tricycloacetoneperoxide Trimeric acetoneperoxide triacetone triperoxide |
| Inchi: | InChI=1S/C9H18O6/c1-7(2)10-12-8(3,4)14-15-9(5,6)13-11-7/h1-6H3 |
| InchiKey: | ZTLXICJMNFREPA-UHFFFAOYSA-N |
| Formula: | C9H18O6 |
| SMILES: | CC1(C)OOC(C)(C)OOC(C)(C)OO1 |
| Mol. weight [g/mol]: | 222.24 |
| CAS: | 17088-37-8 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|--------|----------------|
| gf | -535.56 | kJ/mol | Joback Method |
| hf | -980.21 | kJ/mol | Joback Method |
| hfus | 35.72 | kJ/mol | Joback Method |
| hvap | 59.56 | kJ/mol | Joback Method |
| log10ws | -2.80 | | Crippen Method |
| logp | 2.053 | | Crippen Method |
| mcvol | 162.030 | ml/mol | McGowan Method |
| pc | 3131.49 | kPa | Joback Method |
| rinpol | 1128.10 | | NIST Webbook |
| rinpol | 1120.60 | | NIST Webbook |
| rinpol | 1122.60 | | NIST Webbook |
| rinpol | 1094.00 | | NIST Webbook |
| rinpol | 1115.00 | | NIST Webbook |
| rinpol | 1115.00 | | NIST Webbook |
| rinpol | 1115.30 | | NIST Webbook |
| rinpol | 1125.80 | | NIST Webbook |
| rinpol | 1105.80 | | NIST Webbook |
| rinpol | 1111.90 | | NIST Webbook |

| | | | | |
|--------|---------|--|----------------------|--|
| rinpol | 1094.00 | | | NIST Webbook |
| tb | 590.76 | | K | Joback Method |
| tc | 831.82 | | K | Joback Method |
| tf | 370.15 | | K | A comparative study on two explosive acetone peroxides |
| tf | 370.15 | | K | THERMOCHEMISTRY OF CYCLIC ACETONE PEROXIDES |
| vc | 0.567 | | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|----------|---------|-----------------|---|
| cpg | 450.57 | J/mol×K | 590.76 | Joback Method |
| cpg | 468.30 | J/mol×K | 630.94 | Joback Method |
| cpg | 485.08 | J/mol×K | 671.11 | Joback Method |
| cpg | 501.23 | J/mol×K | 711.29 | Joback Method |
| cpg | 517.04 | J/mol×K | 751.46 | Joback Method |
| cpg | 532.84 | J/mol×K | 791.64 | Joback Method |
| cpg | 548.93 | J/mol×K | 831.82 | Joback Method |
| psub | 6.82e-03 | kPa | 303.15 | Triacetone triperoxide thermogravimetric study of vapor pressure and enthalpy of sublimation in 303 338 K temperature range |
| psub | 0.01 | kPa | 308.15 | Triacetone triperoxide thermogravimetric study of vapor pressure and enthalpy of sublimation in 303 338 K temperature range |
| psub | 0.02 | kPa | 313.15 | Triacetone triperoxide thermogravimetric study of vapor pressure and enthalpy of sublimation in 303 338 K temperature range |

| | | | | |
|------|------|-----|--------|---|
| psub | 0.03 | kPa | 318.15 | Triacetone triperoxide thermogravimetric study of vapor pressure and enthalpy of sublimation in 303 338 K temperature range |
| psub | 0.05 | kPa | 323.15 | Triacetone triperoxide thermogravimetric study of vapor pressure and enthalpy of sublimation in 303 338 K temperature range |
| psub | 0.07 | kPa | 328.15 | Triacetone triperoxide thermogravimetric study of vapor pressure and enthalpy of sublimation in 303 338 K temperature range |
| psub | 0.10 | kPa | 333.15 | Triacetone triperoxide thermogravimetric study of vapor pressure and enthalpy of sublimation in 303 338 K temperature range |
| psub | 0.12 | kPa | 338.15 | Triacetone triperoxide thermogravimetric study of vapor pressure and enthalpy of sublimation in 303 338 K temperature range |

Sources

Triacetone triperoxide thermogravimetric study of vapor pressure and enthalpy of sublimation in 303 338 K temperature range: THERMOCHEMISTRY OF CYCLIC ACETONE PEROXIDES: Joback Method:

<https://www.doi.org/10.1016/j.tca.2010.11.034>

<https://www.doi.org/10.1016/j.tca.2013.08.009>

<https://www.doi.org/10.1016/j.tca.2014.03.046>

https://en.wikipedia.org/wiki/Joback_method

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C17088378&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| psub: | Sublimation pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

Latest version available from:

<https://www.chemeo.com/cid/20-993-7/TATP.pdf>

Generated by Cheméo on 2024-04-23 13:53:43.590907475 +0000 UTC m=+16169672.511484806.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.