

# 1,1,1-Tris(hydroxymethyl)propane, tri(trifluoroacetate)

Inchi:	InChI=1S/C12H11F9O6/c1-2-9(3-25-6(22)10(13,14)15,4-26-7(23)11(16,17)18)5-27-8(24)
InchiKey:	AEKOLUVECIUHSH-UHFFFAOYSA-N
Formula:	C12H11F9O6
SMILES:	CCC(COC(=O)C(F)(F)F)(COC(=O)C(F)(F)F)COC(=O)C(F)(F)F
Mol. weight [g/mol]:	422.20

## Physical Properties

Property code	Value	Unit	Source
gf	-2393.53	kJ/mol	Joback Method
hf	-2825.40	kJ/mol	Joback Method
hfus	33.26	kJ/mol	Joback Method
hvap	57.24	kJ/mol	Joback Method
log10ws	-3.18		Crippen Method
logp	2.699		Crippen Method
mcvol	218.190	ml/mol	McGowan Method
pc	1524.69	kPa	Joback Method
rinsol	1098.00		NIST Webbook
tb	683.34	K	Joback Method
tc	847.33	K	Joback Method
tf	456.47	K	Joback Method
vc	0.897	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	645.46	J/mol×K	683.34	Joback Method
cpg	656.04	J/mol×K	710.67	Joback Method
cpg	665.89	J/mol×K	738.00	Joback Method
cpg	675.06	J/mol×K	765.33	Joback Method
cpg	683.57	J/mol×K	792.67	Joback Method
cpg	691.47	J/mol×K	820.00	Joback Method
cpg	698.78	J/mol×K	847.33	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U374842&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U374842&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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