

# Hexane, 2,3,3,4-tetramethyl-

<b>Other names:</b>	2,3,3,4-Tetramethylhexane
<b>Inchi:</b>	InChI=1S/C10H22/c1-7-9(4)10(5,6)8(2)3/h8-9H,7H2,1-6H3
<b>InchiKey:</b>	HIHSOGFAVTVMCY-UHFFFAOYSA-N
<b>Formula:</b>	C10H22
<b>SMILES:</b>	CCC(C)C(C)(C)C(C)C
<b>Mol. weight [g/mol]:</b>	142.28
<b>CAS:</b>	52897-10-6

## Physical Properties

Property code	Value	Unit	Source
gf	31.28	kJ/mol	Joback Method
hf	-269.04	kJ/mol	Joback Method
hfus	7.20	kJ/mol	Joback Method
hvap	46.40	kJ/mol	NIST Webbook
log10ws	-3.28		Crippen Method
logp	3.715		Crippen Method
mcvol	151.760	ml/mol	McGowan Method
pc	2177.49	kPa	Joback Method
rinpol	949.00		NIST Webbook
rinpol	949.00		NIST Webbook
rinpol	949.00		NIST Webbook
rinpol	946.00		NIST Webbook
rinpol	949.00		NIST Webbook
rinpol	949.00		NIST Webbook
rinpol	946.00		NIST Webbook
rinpol	949.10		NIST Webbook
tb	437.74 ± 0.20	K	NIST Webbook
tc	604.51	K	Joback Method
tf	174.88	K	Joback Method
vc	0.573	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	316.88	J/molxK	424.09	Joback Method
cpg	334.37	J/molxK	454.16	Joback Method
cpg	351.04	J/molxK	484.23	Joback Method
cpg	366.91	J/molxK	514.30	Joback Method
cpg	382.02	J/molxK	544.37	Joback Method
cpg	396.39	J/molxK	574.44	Joback Method
cpg	410.05	J/molxK	604.51	Joback Method
dvisc	0.0446565	Paxs	174.88	Joback Method
dvisc	0.0079881	Paxs	216.41	Joback Method
dvisc	0.0024872	Paxs	257.95	Joback Method
dvisc	0.0010703	Paxs	299.49	Joback Method
dvisc	0.0005656	Paxs	341.02	Joback Method
dvisc	0.0003433	Paxs	382.55	Joback Method
dvisc	0.0002298	Paxs	424.09	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.40574e+01
Coeff. B	-3.59768e+03
Coeff. C	-5.66020e+01
Temperature range (K), min.	317.88
Temperature range (K), max.	467.96

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C52897106&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C52897106&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/21-215-9/Hexane-2-3-3-4-tetramethyl.pdf>

Generated by Cheméo on 2024-04-26 17:52:52.651207701 +0000 UTC m=+16443221.571785013.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.