

# Diethylmalonic acid, monochloride, 2-methoxyethyl ester

<b>Inchi:</b>	InChI=1S/C10H17ClO4/c1-4-10(5-2,8(11)12)9(13)15-7-6-14-3/h4-7H2,1-3H3
<b>InchiKey:</b>	UPMSXHDZLSIVJZ-UHFFFAOYSA-N
<b>Formula:</b>	C10H17ClO4
<b>SMILES:</b>	CCC(CC)(C(=O)Cl)C(=O)OCCOC
<b>Mol. weight [g/mol]:</b>	236.69

## Physical Properties

Property code	Value	Unit	Source
gf	-443.61	kJ/mol	Joback Method
hf	-763.82	kJ/mol	Joback Method
hfus	24.01	kJ/mol	Joback Method
hvap	59.25	kJ/mol	Joback Method
log10ws	-1.64		Crippen Method
logp	1.748		Crippen Method
mcvol	178.880	ml/mol	McGowan Method
pc	2258.96	kPa	Joback Method
rinpol	1354.00		NIST Webbook
rinpol	1354.00		NIST Webbook
tb	614.98	K	Joback Method
tc	808.91	K	Joback Method
tf	379.12	K	Joback Method
vc	0.681	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	446.78	J/molxK	614.98	Joback Method
cpg	505.40	J/molxK	776.59	Joback Method
cpg	495.02	J/molxK	744.27	Joback Method
cpg	483.98	J/molxK	711.95	Joback Method
cpg	472.27	J/molxK	679.62	Joback Method
cpg	459.88	J/molxK	647.30	Joback Method
cpg	515.12	J/molxK	808.91	Joback Method
dvisc	0.0001563	Paxs	614.98	Joback Method

dvisc	0.0002040	Paxs	575.67	Joback Method
dvisc	0.0002769	Paxs	536.36	Joback Method
dvisc	0.0003945	Paxs	497.05	Joback Method
dvisc	0.0005973	Paxs	457.74	Joback Method
dvisc	0.0009776	Paxs	418.43	Joback Method
dvisc	0.0017721	Paxs	379.12	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370686&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370686&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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