

Sarcosine, N-(2,3,4-trifluorobenzoyl)-, isoheptyl ester

Inchi:	InChI=1S/C16H20F3NO3/c1-10(2)5-4-8-23-13(21)9-20(3)16(22)11-6-7-12(17)15(19)14(18)
InchiKey:	ZHIGJQSJWMSFAF-UHFFFAOYSA-N
Formula:	C16H20F3NO3
SMILES:	CC(C)CCCOC(=O)CN(C)C(=O)c1ccc(F)c(F)c1F
Mol. weight [g/mol]:	331.33

Physical Properties

Property code	Value	Unit	Source
gf	-671.57	kJ/mol	Joback Method
hf	-1054.91	kJ/mol	Joback Method
hfus	43.19	kJ/mol	Joback Method
hvap	70.58	kJ/mol	Joback Method
log10ws	-4.16		Crippen Method
logp	3.155		Crippen Method
mcvol	236.840	ml/mol	McGowan Method
pc	1615.47	kPa	Joback Method
rinpol	2049.00		NIST Webbook
rinpol	2049.00		NIST Webbook
tb	747.07	K	Joback Method
tc	934.44	K	Joback Method
tf	475.39	K	Joback Method
vc	0.919	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	683.23	J/mol×K	747.07	Joback Method
cpg	697.23	J/mol×K	778.30	Joback Method
cpg	710.39	J/mol×K	809.53	Joback Method
cpg	722.71	J/mol×K	840.75	Joback Method
cpg	734.23	J/mol×K	871.98	Joback Method
cpg	744.97	J/mol×K	903.21	Joback Method
cpg	754.93	J/mol×K	934.44	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321478&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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