

Diethylmalonic acid, 2-fluorophenyl heptadecyl ester

Inchi:	InChI=1S/C30H49FO4/c1-4-7-8-9-10-11-12-13-14-15-16-17-18-19-22-25-34-28(32)30(5-
InchiKey:	IWRDUVBECOFNTL-UHFFFAOYSA-N
Formula:	C30H49FO4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccccc1F
Mol. weight [g/mol]:	492.71

Physical Properties

Property code	Value	Unit	Source
gf	-355.31	kJ/mol	Joback Method
hf	-1131.93	kJ/mol	Joback Method
hfus	68.35	kJ/mol	Joback Method
hvap	101.51	kJ/mol	Joback Method
log10ws	-9.95		Crippen Method
logp	8.952		Crippen Method
mcvol	426.450	ml/mol	McGowan Method
pc	724.18	kPa	Joback Method
rinsol	3215.00		NIST Webbook
tb	1066.08	K	Joback Method
tc	1317.90	K	Joback Method
tf	614.13	K	Joback Method
vc	1.663	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1503.29	J/mol×K	1066.08	Joback Method
cpg	1522.47	J/mol×K	1108.05	Joback Method
cpg	1539.84	J/mol×K	1150.02	Joback Method
cpg	1555.53	J/mol×K	1191.99	Joback Method
cpg	1569.65	J/mol×K	1233.96	Joback Method
cpg	1582.35	J/mol×K	1275.93	Joback Method
cpg	1593.76	J/mol×K	1317.90	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U370141&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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