

# Propanedioic acid, acetyl-, diethyl ester

<b>Other names:</b>	Malonic acid, acetyl-, diethyl ester C-Acetylmalonic ester Diethyl acetylmalonnate Ethyl acetomalonate Diethyl 2-acetylmalonate Ethyl acetylmalonate Diethyl acetomalonate
<b>Inchi:</b>	InChI=1S/C9H14O5/c1-4-13-8(11)7(6(3)10)9(12)14-5-2/h7H,4-5H2,1-3H3
<b>InchiKey:</b>	SQAUUQRBOCJRCW-UHFFFAOYSA-N
<b>Formula:</b>	C9H14O5
<b>SMILES:</b>	CCOC(=O)C(C(C)=O)C(=O)OCC
<b>Mol. weight [g/mol]:</b>	202.20
<b>CAS:</b>	570-08-1

## Physical Properties

Property code	Value	Unit	Source
gf	-574.30	kJ/mol	Joback Method
hf	-836.55	kJ/mol	Joback Method
hfus	22.72	kJ/mol	Joback Method
hvap	60.30	kJ/mol	Joback Method
log10ws	-0.35		Crippen Method
logp	0.318		Crippen Method
mcvol	154.120	ml/mol	McGowan Method
pc	2715.50	kPa	Joback Method
tb	611.33	K	Joback Method
tc	804.41	K	Joback Method
tf	370.44	K	Joback Method
vc	0.588	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	385.35	J/mol×K	611.33	Joback Method
cpg	397.13	J/mol×K	643.51	Joback Method

cpg	408.34	J/molxK	675.69	Joback Method
cpg	418.99	J/molxK	707.87	Joback Method
cpg	429.06	J/molxK	740.05	Joback Method
cpg	438.55	J/molxK	772.23	Joback Method
cpg	447.44	J/molxK	804.41	Joback Method
dvisc	0.0020254	Paxs	370.44	Joback Method
dvisc	0.0011397	Paxs	410.59	Joback Method
dvisc	0.0007104	Paxs	450.74	Joback Method
dvisc	0.0004785	Paxs	490.88	Joback Method
dvisc	0.0003421	Paxs	531.03	Joback Method
dvisc	0.0002564	Paxs	571.18	Joback Method
dvisc	0.0001996	Paxs	611.33	Joback Method
hvapt	54.00	kJ/mol	436.50	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	398.00	K	2.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C570081&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C570081&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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